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Synthesis of the JKLM-ring fragment of ciguatoxin

Takayuki Baba, Guobin Huang and Minoru Isobe*

Laboratory of Organic Chemistry, Graduate School of Bioagricultural Sciences, Nagoya University, Chikusa, Nagoya 464-8601, Japan

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Abstract—A stereoselective synthesis of the LM-ring fragment has been achieved starting from a sugar derivative. A stereoselective synthesis of the JKLM-ring fragment has been achieved through a coupling between two segments via heteroconjugate addition, sevenmembered ether ring formation mediated by an acetylene cobalt complex, and spiroketalization reaction. Q 2003 Elsevier Ltd. All rights reserved.

1. Introduction

Ciguatoxin (CTX) is a principal toxin of ciguatera, which is known as the most widespread seafood poisoning.^{[1](#page-19-0)} The causative toxins of this poisoning produced by the epiphytic dinoflagellate, Gambierdiscus toxicus, [2](#page-19-0) are accumulated in carnivorous fish of many species through the food chain among coral biota, and finally causing human intoxication. The poisoning symptom does crisis to more than 20000 people annually in the world.^{[1](#page-19-0)} It is a serious problem especially in the societies of tropical and subtropical regions. CTX was first isolated from moray eel, Gymnothorax javanicus, by Scheuer and co-workers at the University of Hawaii and characterized as a polyether compound in 1980.^{[3](#page-19-0)} Thus far, more than 23 congeners of CTX have been identified to date.^{[1](#page-19-0)} Ciguatoxins (CTXs) and another structurally related marine toxin, brevetoxins (BTXs), are selective sodium channel activators, which bind quasi-irreversibly to site 5 on the voltage-sensitive sodium channels (VSSC) in nerves, heart and muscle.^{[4,5](#page-19-0)} In spite of structural similarity to BTXs, the binding affinity of CTX was shown to be some ten times more potent than that of BTXs.[4](#page-19-0) CTX remains the most potent neurotoxin known with a mouse lethality LD_{50} of 0.35 μ g/kg (i.p.).^{[4](#page-19-0)}

Since Yasumoto and co-workers determined the gross structure of CTX in 1989 ,^{[6](#page-20-0)} its complicated structure has been in the foreground of attention among the scientists. The chemical construction of CTX is a trans-fused polycyclic system composed of a single carbon chain that winds the length of the molecule and linking by ether oxygens into a series of five- to nine-membered oxacycles. Its absolute configuration was successfully elucidated by Yasumoto and co-workers in 1997 as shown in [Figure 1](#page-1-0).^{[7](#page-20-0)}

Several synthetic groups have been studying the total synthesis of CTX over last decade.^{[8](#page-20-0)} Recently, Hirama's group reported the first total synthesis of CTX3C, a member of the CTX family.^{[9](#page-20-0)} We also have endeavored to develop effective methodologies, and established valid methodologies for the construction of medium-sized ether rings via cobalt complex-mediated cyclization during the course of our studies toward the synthesis of CTX.[10](#page-20-0) We have already achieved the syntheses of the ABC rings with the side chain,^{[11](#page-20-0)} the BCDE,^{[12](#page-20-0)} the D'EF,^{[13](#page-20-0)} the E'FGH^{[14](#page-20-0)} rings^{[15](#page-20-0)} using acetylene cobalt complex strategy. With regard to right part of CTX, we previously reported a model study of stereoselective synthesis of the H'IJK-ring fragment.^{[16](#page-20-0)} In this paper, we provide full detail of the synthesis of the LMand the JKLM-ring fragments as the right end of CTX.

2. Synthesis of the LM-ring fragment

Firstly, we set about the synthesis of the LM-ring fragment of CTX.[17](#page-20-0) Its retrosynthetic analysis is shown in [Scheme 1](#page-1-0). We anticipated that LM-ring system could be derived from hemiacetal 3 by asymmetric dihydroxylation reaction and then cyclization. Hemiacetal 3 could be prepared from α, β unsaturated lactone 4 through a stereocontrolled conjugate addition and enolate trapping reaction. Lactone 4 would be synthesized starting from tri-O-acetyl-D-galactal 5.

The synthesis of the initial target 1 is outlined in [Scheme 2](#page-2-0). $Tri-O$ -acetyl-D-galactal 5 was deacetylated under the condition of NaOMe/MeOH to afford D-galactal 6 in 87% yield. The primary hydroxyl group of $\vec{6}$ was selectively silylated, and the allylic hydroxyl group was then selectively protected by benzoylation with benzoyl chloride and pyridine under -35° C. The remaining hydroxyl group was protected as the benzyl ether to afford 9. After desilylation with TBAF and protection with TBDPS, the protected D-galactal was oxidized by PCC at 80° C to give

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^{*} Corresponding author. Tel.: $+81-52-789-4109$; fax: $+81-52-789-4111$; e-mail: isobem@agr.nagoya-u.ac.jp

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lactone $12¹⁸$ $12¹⁸$ $12¹⁸$ which was treated with DBU to give the desired enlactone 13 in 98% yield.^{[19](#page-20-0)} With the key enlactone 13 in hand, it was treated with $Me₂CuLi$ to give monomethylated lactone in 91% yields. After the monomethylated lactone was treated with LiHMDS for 30 min at -78° C, MeI was added to the reaction mixture at this temperature to give dimethyllactone 14 and its C51-epimer with the ratio of 7:1 in 93% combined yield. Addition of allylmagnesium bromide to dimethyllactone 14 provided hemiacetal 15. Asymmetric dihydroxylation and spiroketal-ization were conducted under Sharpless condition^{[20](#page-20-0)} to afford a mixture of four isomers 16a–d with the ratio of 4:4:1:1 in 75% combined yield, which were easily separated by preparative TLC.

The stereochemistry of 16a–d was determined through careful analysis of their ¹H NMR and NOESY spectra, shown as following. The important data of compound 16a are the coupling constants $J_{50,51}$ =11.0 Hz, $J_{49,50}$ =11.0 Hz, and the observation of the cross peaks between H-53 α and $H-54$, $H-53\beta$ and C60–Me on its NOESY experiment. These data indicate that the conformation of the pyranose-nucleus and the stereochemistry of C54 are as depicted in [Figure 2.](#page-2-0) For 16b, the coupling constants $J_{50,51} = 2.0$ Hz, $J_{49,50}$ =1.5 Hz and the observation of the cross peak between H-48 and C59–Me on its NOESY experiment suggested that C59-Me and C60-Me are axial. The observation of the cross peaks between H-53 β and H-54, H-53 β and C60–Me indicated that the stereochemistry of its C54 is S configuration.

Scheme 1. Retrosynthetic analysis of the LM-ring fragment 1.

The stereochemistry of compounds **16c** and **d** was also determined through careful analysis of their NMR spectra, shown in [Figure 2](#page-2-0). The stereochemistry at C54 in 16c and d could be easily inverted into 16b and 1a by Mitsunobu reaction, 21 respectively. The mixture of 16c and d were treated with DEAD, PPh_3 and p-nitrobenzoic acid, then treated with $K_2CO_3/MeOH$ to provide a mixture of 16c and b in 95% yield [\(Scheme 3\)](#page-2-0).

3. Synthesis of the JKLM-ring fragment

3.1. Retrosynthetic analysis

Having accomplished the synthesis of the LM-ring fragment, we now could consider the synthesis of the JKLM-ring fragment of CTX. According to our plan, the synthesis of CTX could be achieved via the coupling of two large segment, acetylene in the Segment L 17 with aldehyde in Segment R 18 ([Scheme 4\)](#page-3-0). This would be followed by the construction of central part (FG-ring), and finally A-ring cyclization. We have already reported the synthesis of Segment L 17.^{[12b](#page-20-0)} The retrosynthetic analysis for the right part of CTX is illustrated in [Scheme 5](#page-3-0). Segment R 18 could be derived from acetylene 19, representing C30–C38 portion of CTX, and the JKLM-ring fragment 2. Opening of the terminal spiroketal in 2 provides 20 as a synthetic equivalent. The seven-membered ring in 20 would be constructed via acetylene cobalt complex 21. Opening of the seven-membered ring K in 21 gives 22, which further leads us to the two segments vinyl sulfone 23 and acetylene 24 to be coupled between the C46 and C47 positions on the basis of a heteroconjugate addition. $10j-1$

3.2. Synthesis of the acetylene subsegment

We attempted to transform the LM-ring system of 16 into acetylene 24 [\(Scheme 6](#page-4-0)). The C54 hydroxyl groups of 16a and b were protected by benzyl group, and then treated with TBAF to afford a mixture of alcohols 25 in 92% yield. The primary alcohol of 25 was successfully converted into chloride and iodide under ordinary conditions to provide 26 and 27. Both of the attempts for the transformation of 26 and 27 into acetylene under basic conditions were, however, unsuccessful.^{[22](#page-20-0)} We have also tried direct opening of

Scheme 2. Reagents, conditions and yields: (a) NaOMe, MeOH, 87%; (b) TBSCl, Py, DMF, room temperature, 68%; (c) BzCl, Py, DMAP, -35°C, 93%; (d) BnBr, NaH, DMF, 92%; (e) TBAF, THF, 95%; (f) TBDPSCl, imidazole, DMF, 100% ; (g) PCC, (CH2Cl)₂, 51%; (h) DBU, CH₂Cl₂, 98%; (i) Me₂CuLi, Et₂O 98%; (j) LiHMDS, MeI, THF, -78°C (14-C51-epimer=7:1); (k) CH₂=CHCH₂MgBr, Et₂O, -78°C 87%; (l) AD-mix-a, t-BuOH, H₂O, 75%.

Scheme 3. Reagents, conditions and yields: (i) p -NO₂C₆H₄CO₂H, DEAD, PPh₃, toluene; (ii) K_2CO_3 , MeOH, 95% in 2 steps.

spiroketal in 25 with 1,3-propanedithiol catalyzed with BF_3 ·OEt₂ to afford dithiane product,^{[23](#page-20-0)} but it could not afford the desired product. To our regret, it would seem no other efficient way to derive the acetylene compound 24 from LM-ring system 16, though we explored every avenue.

Faced with this impasse, we were forced to abandon this line, and seek out another way. An alternative strategy for construction of the acetylene segment 24 is shown in [Scheme 7.](#page-4-0) Thus, tri-O-acetyl-D-glucal was converted to the enone 31 by a four-step sequence; O-glycosidation with

Scheme 4.

2-propanol catalyzed by BF_3 ·OEt₂, saponification with basic MeOH, silylation under the condition of TBSCl/imidazole and oxidation by $DMSO/Ac₂O$. These steps were amenable to a large-scale operation. 1,4-Addition of lithium dimethyl cuprate to α , β -unsaturated carbonyl of 31, followed by enolate trapping with MeI in the presence of N , N dimethylacetamide as a co-solvent provided 32 as an exclusive diastereomer. The stereochemistry of 32 was confirmed by the analysis of its data of ${}^{1}H$ NMR and NOESY experiment ([Fig. 3](#page-4-0)). The carbonyl group was stereoselectively reduced to the alcohol by $NabH(OAc)₃²⁴$ $NabH(OAc)₃²⁴$ $NabH(OAc)₃²⁴$ after the removal of the TBS group of 32 to afford the diol 34. Opening of the pyranose ring of 34 with 1,3 propanedithiol was unsuccessful under BF_3 ·OEt₂ catalyzed condition. On the other hand, concentrated hydrochloric acid in chloroform provided open-chain triol compound in nearly quantitative yield, which was subsequently protected with TBS and isopropylidene group to give 37. Coupling reaction of the lithio derivative of dithiane 37 with glycidyl methoxybenzyl ether proceeded uneventfully under mild condition 25 and afforded an alcohol, which was protected with benzyl group together with the primary alcohol after desilylation. Subsequent acidic hydrolysis of the isopropylidene group afforded 41. Oxidative cleavage of the 1,2-diol 41 by $Pb(OAc)₄$ provided the corresponding aldehyde 42. The aldehyde was treated with lithium TMS acetylide and MeI, and then desilylated with TBAF to give the acetylene 44. Finally, removal of the dithiane group was performed by brief treatment of 44 with N-chlorosuccinimide and $AgNO₃$ in wet acetonitrile containing $2,4,6$ -collidine.^{[26](#page-20-0)} The yield of this reaction was moderate probably due to the instability of acetylene moiety under the reaction condition, though this was the most suitable method for unmasking of the ketone group of 44. Several other procedures were also tested for this conversion (e.g. CuCl₂ and CuO in wet acetone, 27 27 27 [bis(trifluoroacetoxy)iodo]-benzene²⁸ or MeI²⁹ in wet acetonitrile), but they produced substantial amount of inseparable byproduct. The unmasked ketone was reduced to an alcohol 46 (diastereomeric mixture at C49; ca. 2:1) which was protected by TBS group to afford the targeted compound 47.

Scheme 5. Retrosynthetic analysis of right part of ciguatoxin.

T. Baba et al. / Tetrahedron 59 (2003) 6851–6872 6855

Scheme 6. Reagents, conditions and yields: (a) (i) BnBr, NaH, DMF, 88%, (ii) TBAF, THF, 92%; (b) CCl₄, PPh₃, 79%; (c) I_2 , imidazole, PPh₃, 82%; (d) LDA, THF, -78° C; (e) DBU, THF, reflux; (f) HS(CH₂)3SH, BF₃·OEt₂, -40° C to room temperature.

3.3. Synthesis of the vinyl sulfone subsegment

The construction of the other requisite subsegment for the JKLM-ring system bearing vinyl sulfonyl group is illustrated in [Scheme 8.](#page-5-0) Synthesis of the vinyl sulfone 69 began from methyl α -D-glucopyranoside derivative 48.^{[12](#page-20-0)} Thus, the hydroxyl group of the C42 position in 48 was selectively protected by pivaloyl group.^{[30](#page-20-0)} The remaining free hydroxyl group in 49 was converted to the thiocarbamate by treating with in situ generated thiocarbonyldiimidazole, and removed under modified Barton conditions^{[31](#page-20-0)} to afford

compound 51. The protective groups in 51 were manipulated to provide 55 in 4 steps under standard conditions. The acetal 55 was transformed to lactone 58 via acetolysis with sulfuric acid in acetic anhydride, hydrolysis with aqueous hydrochloric acid in ethylene glycol dimethyl ether and oxidation of the anomeric position. Addition of allylmagnesium bromide to the lactone 58, followed by Kishi's silane reduction, 32 provided hydropyran system 59 as an exclusive diastereomer. Removal of the pivaloyl group from 59 led to diol 60 which was converted to 62 through disilylation and selective removal of the silyl group attached to the primary hydroxyl group.[33](#page-21-0) Oxidation of the primary alcohol followed by dibromo-olefination of the resulting aldehyde 63 gave the vinyl dibromide $64³⁴$ $64³⁴$ $64³⁴$ which was converted to the thiophenylacetylene 65 by further treatment with n -BuLi and PhSSO₂Ph. Then the TBS group was exchanged to acetyl group to give 66. Concordant with our previous work, this thiophenylacetylene underwent highly regioselective hydrosilylation in the presence of a catalytic amount of cobalt complex to afford the corresponding vinylsilane 67, [35](#page-21-0) albeit minor amount of an inseparable isomer (later determined to be the inner olefin isomer of allyl group) could be detected by ${}^{1}H$ NMR. In this reaction, stoichiometric use of the cobalt complex caused increase of isomerization of terminal olefin into inner olefin. And we found that the isomerization is due to the activity of $Co₂(CO)₆$ species liberated from the catalyst. However, to our delight, the minor isomer gradually filtered out over the course of the remainder of the synthesis. Finally, removal of

Scheme 7. Reagents, conditions and yields: (a) i-PrOH, BF₃·OEt₂, CH₂Cl₂; (b) Et₃N, MeOH, H₂O, 84% in 2 steps; (c) TBSCl, imidazole, DMF; (d) Ac₂O, DMSO, 97% in 2 steps; (e) CuI, MeLi, Et₂O, 0°C, then MeI, DMA, 92%; (f) TBAF, THF, 82%; (g) NaBH(OAc)₃, CH₃CN, AcOH, 93%; (h) 1,3-propanedithiol, HCl, CHCl₃; (i) TBSCl, Et₃N, DMAP, CH₂Cl₂, 89% in 2 steps; (j) 2,2-dimethoxypropane, PPTS, CH₂Cl₂, quant.; (k) t-BuLi, (2S)-glycidylmethoxybenzyl ether, THF, HMPA, 96%; (1) TBAF, THF; (m) NaH, BnBr, DMF; (n) 80% AcOH, 70% in 3 steps; (o) Pb(OAc)₄, CH₂Cl₂, 99%; (p) n-BuLi, TMS–acetylene, THF, then MeI; (q) TBAF, THF, 86% in 2 steps; (r) NCS, AgNO₃, 2,4,6-collidine, CH₃CN, H₂O; (s) NaBH₄, MeOH; (t) TBSOTf, Py, CH₃CN, 54% in 3 steps.

Scheme 8. Reagents, conditions and yields: (a) PivCl, Py, CH₂Cl₂, 68%; (b) thiophosgene, imidazole, CHCl₃, toluene, 90°C; (c) AIBN, NaH₂PO₂, 2-methoxyethanol, reflux, 87% in 2 steps; (d) NaOMe, MeOH, 80%; (e) KOH, BnCl; (f) Amberlyst 15E®, MeOH, 86% in 2 steps; (g) PivCl, Py, CH₂Cl₂; (h) H₂SO₄, Ac₂O, 96% in 2 steps; (i) HCl, DME, H₂O, 63%; (j) Ac₂O, DMSO, 98%; (k) (i) CH₂=CHCH₂MgBr, THF, -78°C, (ii) Et₃SiH, BF₃·OEt₂, CH₃CN, -10°C, 66% in 2 steps; (l) NaOMe, MeOH, 93%; (m) TBSCl, imidazole, DMF; (n) CSA, MeOH, 88% in 2 steps; (o) (ClCO)2, DMSO, CH2Cl2; (p) CBr4, PPh3, CH_2Cl_2 , 91% in 2 steps; (q) n-BuLi, THF, -78 to 0°C, then PhSSO₂Ph; (r) (i) TBAF, THF, (ii) Ac₂O, Py, 77% in 3 steps; (s) Et₃SiH, biscobalthexacarbonyl-2methyl-but-3-yn-2-ol (cat.), $60^{\circ}C$, (CH_2Cl) ; (t) K₂CO₃, MeOH; (u) mCPBA, Na₂HPO₄,CH₂Cl₂, 85% in 3 steps.

the acetyl group, followed by treatment with $mCPBA$ in the presence of sodium hydrogen phosphate provided the vinyl sulfone 69.

3.4. Heteroconjugate addition and K ring cyclization

Having accomplished the preparation of both subsegments 46 and 47 for the elaboration of the JKLM-ring system, our attention turned to the coupling of these two compounds. The coupling between 47 and 69 and subsequent K ring cyclization are depicted in [Scheme 9.](#page-6-0) To our temporary delight, application of the condition employed in our previous model studies produced a coupling compound in good vield.^{[16a,b](#page-20-0)} Thus, generation of the lithium acetylide of 47 with *n*-BuLi in THF, followed by addition of 69 , gave diastereomeric mixture of 70 in 80% yield. However, NMR studies with 70 were ambiguous, and our empirical method for determination of the stereochemistry of adduct established through related heteroconjugate additions was not sufficiently consistent to allow us to assign with confidence the C46 configuration of 70. While the stereochemical assignments were tentative at this point, the crucial cyclization reaction of the K ring was studied.

We are aware that some attempts for the cyclization of the K ring need rather concentrated Lewis acid condition (around

10 times stronger) as compared with homologous medium sized ring formation we previously conducted using acetylene cobalt complex. $11-15$ In addition, functional group at C52 turned to be a dominant factor in this reaction; attempted formation of the K ring from 1,3-dithiane derivative or TIPS ether at C52, for example, failed due to competing nucleophilic participation of the heteroatom on the functional group at C52. Therefore, the C52 TBS group, having served its role in the coupling reaction, was now replaced by an acetyl group for the purpose of diminishing the nucleophilicity at the fifth position from the cationic center. Application of the usual methods for exchange of protective groups to the above adduct 70 delivered the expected acetate 72, which could be readily converted into corresponding acetylene cobalt complex 73 by simply mixing with $Co_2(CO)_8$ in CH₂Cl₂. Upon treatment of 73 with BF_3 ·OEt₂, the K ring cyclization took place with attendant loss of the PMB group attached to the primary hydroxyl group to afford the bicyclic compound 74. Reductive decomplexation of 74 was conducted with an excess amount of Bu_3SnH under heating in toluene.^{[36](#page-21-0)} This reaction provided the corresponding endocyclic olefin together with an inseparable mixture of the inner olefin isomers of allyl group, similar to hydrosilylation of thiophenylacetylene 66 in Scheme 8, but this time the amount of the undesired olefin isomer was considerable.

T. Baba et al. / Tetrahedron 59 (2003) 6851–6872 6857

Scheme 9. Reagents, conditions and yields: (a) (i) n-BuLi, THF, (ii) TBAF, THF, 87% in 2 steps; (b) ethyl vinyl ether, PPTS, CH₂Cl₂, quant.; (c) (i) TBAF, THF, 88%, (ii) Ac₂O, DMAP, Py, (iii) CSA, MeOH, quant. in 2 steps; (d) Co₂(CO)₈, CH₂Cl₂; (e) BF₃·OEt₂, CH₂Cl₂, 93% in 2 steps; (f) bis-(trimethylsilyl)acetylene, Bu3SnH, toluene, 87% (75: 68%, 76: 19%).

Working with closely related model systems, a reliable protocol was developed for evasion from encumbering isomerization during the reductive decomplexation with hydrosilylation. By simply adding an excess amount of bis- (trimethylsilyl)acetylene in the decomplexation reaction, the formation of the side product, which stems most likely from the activity of $Co_2(CO)_6$ species liberated from the substrate, could completely suppressed.^{[37](#page-21-0)} Application of this procedure to the problem at hand was quite successful and cleanly afforded the desired endocyclic olefin 75 and its C46-epimer 76, both having syn stereochemistry between H44 and H49, as chromatographically separable products in 87% combined yield $(75-76=3.6:1)$. Thus, it seemed to indicate unequivocally that the stereochemical problem lay simply at C46.

On the basis of our previous work,^{[16a,b](#page-20-0)} it appeared that the heteroconjugate addition reaction of lithium acetylide of 47 (R=Li) toward vinyl sulfone 69 with non-protected β hydroxyl group, which would be well-positioned to direct the addition of nucleophile to the same face of the vinyl sulfone, might proceed with extremely high stereoselectivity under adequate condition. The model studies also showed that the stereoselectivity of heteroconjugate addition is highly dependent on solvent and coordination ability of metal. What was puzzling, however, was that solvent modification had surprisingly little effect on the stereoselectivity in this particular system (Table 1).

3.5. Spiroketalization

With the requisite bicyclic compound 75 available, the only issue that remained was the crucial spiroketalization. The final stage of the synthesis of the JKLM-ring fragment is illustrated in [Scheme 10.](#page-7-0) Removal of the acetyl group in 75, followed by selective protection of the primary alcohol by TBS group and oxidation with IBX, 38 furnished ketone 79. The terminal olefin in 79 was oxidized for the purpose of

differentiation from endocyclic olefin to give methyl ketone 80. The stereoselective dihydroxylation of the endocyclic olefin in 80 was achieved under Sharpless condition 39 to afford 81 and 82 as an equilibrium mixture, which underwent desilylation and spiroketalization in the expected sense by treatment with HF·pyridine in acetonitrile to afford the tetracyclic compound 83 as a major product with a minor spiro-isomer. Finally, reduction with sodium-amalgam in methanol gave the desulfonylation product 84. The stereochemistry of 84 was confirmed through the NOE experiments. The results are shown in [Figure 4](#page-7-0) with arrows.

4. Conclusion

We have achieved an efficient synthesis of the JKLM ring fragment in 16 steps from acetylene 47 and vinyl sulfone 69 based on the convergent strategy. It proceeds with modest stereochemical control at C46, but virtually complete

Table 1. Stereoselectivity at C46 position in coupling between 47 and 69

Entry	Conditions	Yield ^a $(\%)$	Ratio ^b 75:76
$\mathbf{1}$	n -BuLi, THF, 0° C, 30 min	80	3.6:1
$\overline{2}$	<i>n</i> -BuLi, Et ₂ O/hexane, 0 to 15 ^o C, 5 h	74	3.4:1
3	n -BuLi, THF/hexane (1/4), 0° C, 2 h	94	3.1:1
$\overline{4}$	MeLi-LiBr, THF, -78 to -30° C, 2 days	74	2.0:1
5	n -BuLi, Et ₂ O, 0 to 25°C, 11 h	90	1.9:1
6	n -BuLi, THF, -20° C, 4 h	85	1.9:1
7	n -BuLi, LiBr, THF, -78 to 25 \degree C, 3 h	D	
8	NaH, <i>n</i> -BuLi, THF, -78 to 0°C, 30 h	D	
9	EtMgBr, THF, -78 to 25° C, 15 h	NR	
10	EtMgBr, Et ₂ O, -78 to 25° C, 15 h	NR	

D—decomposed; NR—no reaction.
^a Yield of adduct 70.

b The ratios of the stereoisomers at C46 position were indirectly established from the ratios of 75 and 76, which were transformed by 5 steps from the coupling product 70.

6858 T. Baba et al. / Tetrahedron 59 (2003) 6851–6872

Scheme 10. Reagents, conditions and yields: (a) K₂CO₃, MeOH, THF; (b) TBSCl, Et₃N, DMAP, CH₂CH₂, 95% in 2 steps; (c) IBX, DMSO, 97%; (d) PdCl₂, CuCl, DMF, H₂O, O₂, 85%; (e) AD-mix-β, CH₃SO₂NH₂, t-BuOH, H₂O; (f) HF·Py, CH₃CN, 72% in 2 steps; (g) Na–Hg, Na₂HPO₄, MeOH, 82%.

Figure 4.

control at all other positions including thermodynamically driven adjustment of syn selective K ring cyclization and final spiroketalization. Further studies toward the synthesis of the right part of CTX along this line are now in progress.

5. Experimental

5.1. General

Infrared (IR) spectra were recorded on a JASCO FT/IR-8300 spectrophotometer or a Paragon 1000 FT-IR spectrometer and are reported in wave number $(cm⁻¹)$. Proton NMR spectra (¹H NMR) were recorded on a Varian Gemini 2000 (300 MHz), a Bruker ARX-400 (400 MHz) or a JEOL L500 (500 MHz). All samples were dissolved in CDCl₃, and chemical shift values are reported in parts per million (ppm) with tetramethylsilane (TMS, δ 0.00) as an internal standard. Data are reported as follows: chemical shift [integrated intensity, multiplicity (s =singlet, d=doublet, t=triplet, q=quartet, qn=quintet, sep=septet, br= broadened, m=multiplet), coupling constant(s) in Hertz, assignment]. The assignment of NMR spectra was largely achieved from COSY spectra. NOESY experiments were

performed with a Bruker ARX-400 (400 MHz) or a JEOL L500 (500 MHz). Carbon NMR spectra $(^{13}C$ NMR) were recorded on a Varian Gemini 2000 (75.4 MHz), a Bruker ARX-400 (100 MHz) or a JEOL L500 (125 MHz) with proton decoupling. Chemical shift values are reported as δ in parts per million (ppm) with $CDCl₃$ (δ 77.0) as an internal standard. Optical rotations were measured on a JASCO DIP-370 digital polarimeter or a JASCO P-1010-TG polarimeter. High-resolution or low-resolution mass spectra were recorded on a Micromass Q-TOF (ESI) or a JEOL JMS-700 spectrometer (FAB and EI), and are reported in m/z . Elemental analyses were performed by Analytical Laboratory at School of Bioagricultural Sciences, Nagoya University. Reactions were monitored by thin-layer chromatography carried out on 0.25 mm silica gel coated glass plates $60F_{254}$ (Cica Merck, Art 1.05715) using UV light as visualizing agent and 7% ethanolic phosphomolybdic acid, or p-anisaldehyde solution as developing agents. Cica Merck silica gel 60 (particle size 0.063–0.2 mm ASTM) was used for open-column chromatography. Unless otherwise noted, non-aqueous reactions were conducted in ovendried $(200^{\circ}C)$ or flame-dried glassware under inert atmosphere. Dry THF was distilled from potassium metal with benzophenone. Anhydrous $Et₂O$ was purchased from Kanto Chemical Co., Inc. in a bottle as Ethyl Ether Anhydrous. Dry CH_2Cl_2 was distilled from CaH_2 under nitrogen atmosphere. BF_3 OEt_2 were distilled from CaH_2 . All other commercially available reagents were used as received. Hyflo-Super-Cel[®] (nacalai tesque) was used as filter aid.

5.1.1. D-Galactal 6. To a solution of tri-O-acetyl-D-galactal 5 (16.1 g, 59 mmol) in dry MeOH was added NaOMe (54 mg), the mixture was stirred for 3 days, then evaporated under reduced pressure to give a crude product (10.0 g). The residue was filtered through a silica gel column to give a syrup (9.00 g). This syrup was crystallized from AcOEt to give a white solid of D-galactal 6 (7.50 g, 87%). 6: mp 99– 100°C, $[\alpha]_D^{16} = -20.6$ ° (c 1.350, MeOH).

5.1.2. TBS-D-galactal 7. To a mixture of D-galactal 6 (7.30 g, 50.0 mmol), pyridine (7.90 g, 100 mmol) and DMAP (0.30 g) in dry DMF (150 mL) was added TBSCl $(7.90 \text{ g}, 52.5 \text{ mmol})$ in three portions over 1 h at 0 \degree C. After stirring for 1 h at 0° C, the mixture was stirred overnight at room temperature, then poured into 5% NaHCO₃ solution and extracted with AcOEt $(X3)$. The combined organic layer was dried over $Na₂SO₄$, filtered and evaporated under reduced pressure to provide crude oil product. The residue was purified by silica gel chromatography (AcOEt– hexane=1:1) to give TBS-D-galactal 7 (8.90 g, 68%) as colorless oil. *Compound* 7. $[\alpha]_0^{16} = +3.94^\circ$ (c 0.815, CHCl₃).
¹H NMR (CDCl₂, 500 MHz) 8.0.12 (6H_s – Si(CH₂). 0.91 ¹H NMR (CDCl₃, 500 MHz) δ 0.12 (6H, s, –Si(CH₃)₂), 0.91 $(9H, s, -SiC(CH_3)_3)$, 2.73 (1H, br s, -OH), 3.18 (1H, br s, -OH), 3.88 (1H, m, H-48), 3.90–3.99 (2H, ddd, $J=10.5, 5.0,$ 4.0 Hz, H-47a, 47b), 4.10 (1H, m, H-49), 4.31 (1H, m, H-50), 4.72 (1H, dt, $J=6.5$, 2.0 Hz, H-51), 6.38 (1H, dd, J=6.5, 1.5 Hz, H-50). ¹³C NMR (CDCl₃, 125 MHz) δ $-5.48, -5.46, 18.3, 25.8, 63.4, 64.2, 66.1, 75.7, 103.1,$ 144.5.

5.1.3. TBS-Bz-D-Galactal 8. To a solution of TBS-Dgalactal $7(6.50 \text{ g}, 25.0 \text{ mmol})$ in dry pyridine (60.0 mL) was added dropwise benzoyl chloride (3.69 g, 26.3 mmol) over 30 min at -35° C under Ar atmosphere. After stirring for 1.5 h at this temperature, the mixture was warmed to 0° C over 15 min, then poured into saturated $NAHCO₃$ solution and extracted with $AcOEt$ (\times 3). The combined organic layer was washed with brine, dried over $Na₂SO₄$, filtered and evaporated under reduced pressure to provide a crude oil product (10.0 g). The residue was purified by silica gel chromatography $(ACOE-Lear)$ to give TBS-Bz-Dgalactal $\mathbf{8}$ (8.50 g, 93%) as a colorless oil. $\mathbf{8}$: [α] $_{\text{D}}^{20}$ =63.8° (*c* 0.890, CHCl₃) ¹H NMR (CDCl₃, 500 MHz) δ 0.11 (6H, s, – $Si(CH_3)_{2}$, 0.91 (9H, s, $-SiC(CH_3)_{3}$), 2.94 (1H, br s, $-OH$), 3.92 (1H, dd, $J=11.0$, 4.0 Hz, H-47a), 4.04 (1H, dd, $J=11.0$, 6.0 Hz, H-47b), 4.06 (1H, dd, $J=6.0$, 4.0 Hz, H-48), 4.42 $(1H, br s, H-49), 4.81$ (1H, dt, J=6.5, 2.0 Hz, H-51), 5.68 $(1H, dd, J=4.5, 2.0 Hz, H=50), 6.53 (1H, dd, J=6.5, 1.5 Hz,$ H-50), 7.40–7.60 (3H, m, aromatic), 8.10 (2H, d, $J=7.3$ Hz, aromatic). ¹³C NMR (CDCl₃, 125 MHz) δ -5.46, 18.3, 25.8 62.9, 64.1, 68.1, 75.9, 98.6, 128.4, 129.8, 129.9, 133.2, 146.1, 166.2.

5.1.4. TBS-Bz-Bn-D-galactal 9. A slurry of 60% NaH (1.25 g) was placed in a 500 mL round bottom flask, and it was washed with hexane $(X2)$. The residual powder was suspended in dry DMF (100 mL). To this suspension were added at 0° C a solution of TBS-Bz-D-galactal 8 (7.77 g, 21.3 mmol) in dry DMF (140 mL) and benzyl bromide $(4.00 \text{ g}, 23.4 \text{ mmol})$ at 0° C under Ar atmosphere. After stirring for 6 h, the mixture was poured into 5% NaHCO₃ solution, extracted with $Et₂O$ (\times 3). The combined organic layer was washed with 5% NaHCO₃ solution, dried over $Na₂SO₄$, filtered and evaporated under reduced pressure to provide the oil product (10.0 g). The residue was purified by silica gel chromatography ($Et₂O$ –hexane=1:10) to give a colorless oil of TBS-Bz-Bn-D-galactal 9 (8.90 g, 92%). 9: $[\alpha]_D^{16}$ = -98.9° (c 1.23, CHCl₃)^TH NMR (CDCl₃, 500 MHz) δ 0.07 (6H, s, –Si(CH₃)₂), 0.91 (9H, s, –SiC(CH₃)₃), 3.89 $(1H, dd, J=10.5, 5.0 Hz, H-47a), 4.01 (1H, dd, J=10.5,$ 7.5 Hz, H-47b), 4.16 (1H, dd, $J=4.0$, 3.5 Hz, H-49), 4.20 $(1H, m, H-48), 4.60$ $(1H, d, J=12.0$ Hz, $-CH₂Ph), 4.78$ $(1H,$

d, $J=12.0$ Hz, $-CH_2Ph$), 4.86 (1H, dd, $J=6.5$, 3.5 Hz, H-51), 5.76 (1H, t, $J=3.5$ Hz, H-50), 6.46 (1H, dd, $J=6.5$, 1.0 Hz, H-50), 7.21–7.58 (8H, m, aromatic), 8.03 (2H, d, J=7.3 Hz, aromatic). ¹³C NMR (CDCl₃, 125 MHz) δ $-5.35, -5.26, 18.4, 25.9, 60.9, 65.7, 70.8, 73.4, 77.4,$ 98.5, 127.7, 127.8, 127.9, 128.3, 128.4, 129.7, 133.1, 137.9, 145.8, 166.2. FAB-MS 477 $[M^+ + Na]^+$.

5.1.5. Bz-Bn-D-galactal 10. To a solution of TBS-Bz-Bn-Dgalactal 9 (3.24 g, 7.14 mmol) in dry THF (20.0 mL) was added dropwise TBAF (1 M solution in THF, 7.20 mL, 7.20 mmol) at room temperature under Ar atmosphere. After stirring for 3 h, the mixture was concentrated to dryness under reduced pressure. The residue was purified by silica gel (Et₂O–hexane=1:4) to give Bz-Bn-D-galactal 10 $(2.30 \text{ g}, 95\%)$ as a colorless oil. Compound 10. $[\alpha]_D^{16}$ = -170.2° (c 1.00, CHCl₃). ¹H NMR (CDCl₃, 500 MHz) δ 1.88 (1H, br s, –OH), 3.88 (1H, dd, J=10.5, 5.0 Hz, H-47a), 4.04 (1H, dd, J=10.5, 8.0 Hz, H-47b), 4.15 (1H, t, J=4.0 Hz, H-49), 4.27 (1H, m, H-48), 4.55 (1H, d, $J=12.0$ Hz, $-CH_2Ph$, 4.79 (1H, d, $J=12.0$ Hz, $-CH_2Ph$), 4.93 (1H, dd, $J=6.5$, 4.0 Hz, H-51), 5.77 (1H, dt, $J=3.5$, 1.5 Hz, H-50), 6.48 (1H, dd, $J=6.5$, 1.0 Hz, H-50), 7.25– 7.60 (8H, m, aromatic), 8.05 (2H, d, $J=7.5$ Hz, aromatic). ¹³C NMR (CDCl₃, 125 MHz) δ 60.8, 64.8, 71.1, 72.9, 76.1, 98.3, 128.0, 128.1, 128.8 129.7, 129.9, 133.2, 137.3, 145.7, 166.2.

5.1.6. TBDPS-Bz-Bn-D-galactal 11. To a mixture of Bz-Bn-D-galactal 10 (1.44 g, 4.24 mmol) and TBDPSCl (1.40 g, 5.09 mmol) imidazole (634 mg, 9.32 mmol) in dry DMF (20.0 mL) was added dropwise at room temperature. After stirring for 2 h, the mixture was poured into saturated NaHCO₃ solution and extracted with AcOEt $(X3)$. The combined organic layer was washed with saturated NaHCO₃ solution and brine, dried over Na₂SO₄, filtered and evaporated under reduced pressure to give a crude oil. The residue was purified by silica gel chromatography $(Et₂O-hexane=1:10)$ to provide TBDPS-Bz-Bn-D-galactal 11 (2.44 g, 100%). Compound 11: $[\alpha]_D^{16} = -55.3^\circ$ (c 0.560, CHCl₃). ¹H NMR (CDCl₃, 500 MHz) δ 1.09 (9H, s, $-SiC(CH₃)₃$, 4.01 (1H, dd, J=11.0, 5.0 Hz, H-47a), 4.12 $(1H, dd, J=11.0, 7.5 Hz, H-47b), 4.18 (1H, t, J=3.5 Hz, H=47b)$ 49), 4.31 (1H, m, H-48), 4.56 (1H, d, $J=12.0$ Hz, $-CH_2Ph$), 4.74 (1H, d, $J=12.0$ Hz, $-CH_2Ph$), 4.81 (1H, dd, $J=6.0$, 3.5 Hz, H-51), 5.73 (1H, t, $J=4.0$ Hz, H-50), 6.34 (1H, d, J=6.0 Hz, H-50), 7.22–7.74 (18H, m, aromatic), 8.03 (2H, d, J=7.5 Hz, aromatic). ¹³C NMR (CDCl₃, 125 MHz) δ 19.2, 26.5, 26.9, 61.4, 65.5, 7.0, 73.2, 77.2, 98.4, 127.66, 127.68, 127.71, 128.3, 128.4, 129.6, 129.68, 129.69, 129.98, 132.9, 133.4, 133.5, 134.8, 135.59, 135.65, 137.8, 145.7, 166.2.

5.1.7. Lactone 12. The mixture of TBDPS-Bz-Bn-Dgalactal 11 $(2.17 \text{ g}, 3.75 \text{ mm})$ and PCC (2.82 g) in 30 mL 1,2-dichloroethane was heated at 80° C for 6 h under Ar atmosphere. After the reaction was completed, the mixture was poured onto a silica gel column and eluted with hexane–Et₂O=4:1 to provide lactone 12 (1.13 g, 51%) as colorless oil. 12: ¹H NMR (CDCl₃, 500 MHz) δ 1.05 (9H, s, $-SiC(CH_3)_{3}$, 3.05 (2H, ddd, J=17.5, 11.0, 8.0 Hz, H-51), 3.86 (1H, dd, $J=10.5$, 5.5 Hz, H-47a), 3.96 (1H, dd, $J=10.5$, 9.0 Hz, H-47b), 4.40 (1H, m, H-48), 4.42 (1H, m, H-49),

4.73 (1H, d, $J=11.0$ Hz, $-CH_2Ph$), 4.80 (1H, d, $J=11.0$ Hz, $-CH_2$ Ph), 5.42 (1H, ddd, $J=11.0$, 8.0, 2.0 Hz, H-50), 7.22– 7.62 (18H, m, aromatic), 7.98 (2H, dd, $J=7.8$, 1.0 Hz, aromatic). ¹³C NMR (CDCl₃, 125 MHz) δ 19.2, 26.9, 32.2, 61.2, 65.8, 69.7, 71.3, 75.1, 79.0, 127.6, 127.8, 127.9, 128.4, 128.6, 129.2, 129.8, 129.9, 130.1, 132.6, 132.7, 133.6, 135.4, 135.5, 137.5, 165.6, 167.8. FAB-MS 617 $[M+Na]^+$, 578 $[M+H]$ ⁺.

5.1.8. Enlactone 13. To a solution of lactone 12 (1.13 g, 1.9 mmol) in dry CH_2Cl_2 (20.0 mL) was added DBU (380 mg, 2.47 mmol) at room temperature under Ar atmosphere. After stirring for 2 h, the mixture was evaporated under reduced pressure and the residue was purified by silica gel chromatography ($Et₂O$ –hexane=1:2) to give enlactone 13 (889 mg, 98%) as a colorless oil. Compound 13: $[\alpha]_D^{16} = -93.1^\circ$ (c 1.030, CHCl₃). ¹H NMR $(CDCl_3, 500 MHz)$ δ 1.04 (9H, s, $-SiC(CH_3)_3$), 3.91 (1H, dd, $J=10.5$, 5.5 Hz, H-47a), 4.12 (1H, dd, $J=10.3$, 8.5 Hz, H-47b), 4.19 (1H, dd, J=5.5, 3.5 Hz, H-49), 4.44 (1H, ddd, $J=8.5, 5.5, 3.5$ Hz, H-48), 4.60 (2H, s, –CH₂Ph), 6.09 (1H, d, $J=10.0$ Hz, H-51), 6.88 (1H, dd, $J=10.0$, 5.5 Hz, H-50), $7.22-7.63$ (15H, m, aromatic). ¹³C NMR (CDCl₃, 125 MHz) ^d 19.2, 26.8, 61.0, 65.8, 71.9, 79.7, 123.9, 127.7, 127.81, 127.83, 128.0, 128.5, 129.8, 129.9, 132.7, 132.9, 135.4, 135.5, 137.5, 142.8, 162.6. FAB-MS 495 [M+Na]⁺, 473 $[M+H]^{+}$.

5.1.9. Dimethyllactone 14. A solution of MeLi (1.14 M in $Et₂O$, 11.8 mL, 13.5 mmol) was added dropwise to a slurry solution of CuI (1.28 g, 6.75 mmol) in dry Et₂O (20 mL) at 0° C under Ar atmosphere. After the mixture was stirred for 10 min, the copper reagent was treated with TMSCl (2.2 mL, 16.9 mmol). After the mixture was cooled at -20° C, a solution of enlactone 13 (795 mg, 1.69 mmol) in dry $Et₂O$ (10 mL) was added, the mixture was stirred overnight at this temperature. After the reaction was completed, the mixture was poured into NH3/NH4Cl solution ($pH=8$) and extracted with Et₂O. The combined organic layer was washed with saturated $NAHCO₃$ solution and brine, dried over $Na₂SO₄$, filtered and evaporated under reduced pressure to provide crude oil product. The residue was purified by silica gel chromatography $(Et₂O₋)$ hexane=1:2) to give a colorless oil (748 mg, 91%). To a solution of LiHMDS (1 M in THF, 4.00 mL, 4.00 mmol) in dry THF (20 mL) was added dropwise a solution of the oil (1.48 g, 3.03 mmol) in dry THF (25 mL) at -78° C under Ar atmosphere. After stirring for 30 min, methyl iodide (1 mL, 15.16 mmol) was added dropwise. After stirring for 30 min, the mixture was quenched by saturated $NH₄Cl$, then warmed to room temperature and extracted with AcOEt. The combined organic layer was washed with brine, dried over $Na₂SO₄$, filtered and evaporated under reduced pressure to provide the crude oil. This crude oil was purified by silica gel chromatography ($Et₂O$ –hexane=1:5) to give dimethyllactone 14 (1.25 g) and its C51-epimer $14'$ (170 mg) in 93% combined yield. *Compound* 14: ¹H NMR (CDCl₃, 500 MHz) δ 1.01 (9H, s, $-SiC(CH_3)_3$), 1.08 (3H, d, $J=7.0$ Hz, CH₃-59), 1.21 (3H, d, $J=6.5$ Hz, CH₃-60), 1.83 (1H, m, H-50), 2.02 (1H, dq, J=9.0, 7.0 Hz, H-51), 3.53 $(1H, dt, J=2.0, 2.0 Hz, H=49), 3.88-3.97 (2H, ddd, J=10.5,$ 7.5, 5.5 Hz, H-47a), 4.42 (1H, td, J=7.5, 2.0 Hz, H-48), 4.39 (1H, d, J=12.0 Hz, $-CH_2Ph$), 4.57 (1H, d, J=12.0 Hz,

 $-CH_2P$ h), 7.18–7.60 (15H, m, aromatic). ¹³C NMR (CDCl₃ 125 MHz) ^d 14.7, 19.2, 19.3, 26.8, 38.6, 40.1, 61.5, 71.1, 77.1, 77.2, 127.5, 127.7, 127.8, 127.8, 128.4, 129.8, 129.9, 132.9, 135.5, 135.5, 137.8, 174.4. FAB-MS 525 $[M+Na]^{+}$, 503 $[M+H]^+$, 14^{\prime} (C51-epimer of 14): ¹H NMR (CDCl₃, 500 MHz) δ 0.96 (3H, d, J=7.5 Hz, CH₃-60), 1.04 (9H, s, $-SiC(CH_3)$ ₃), 1.18 (3H, d, J=7.0 Hz, CH₃-59), 2.37 (1H, m, H-50), 3.05 (1H, dq, $J=7.5$, 5.0 Hz, H-51), 3.73 (1H, t, J=3.0 Hz, H-49), 3.82 (1H, dd, J=10.0, 5.5 Hz, H-47a), 4.03 (1H, dd, $J=10.0$, 8.5 Hz, H-47b), 4.47 (1H, ddd, $J=8.5$, 5.5, 3.0 Hz, H-48), 4.57 (1H, d, $J=11.4$ Hz, $-CH_2Ph$), 4.68 $(1H, d, J=11.4 \text{ Hz}, -CH₂Ph), 7.24-7.62 (15H, mi, aromatic).$ ¹³C NMR (CDCl₃, 125 MHz) δ 13.0, 13.1, 19.2, 26.8, 34.2, 34.7, 61.7, 71.7, 75.3, 78.6, 127.4, 127.7, 127.8, 128.4, 129.8, 129.9, 132.9, 133.1, 135.5, 135.6, 137.9, 173.5. FAB-MS 525 $[M+Na]^{+}$, 503 $[M+H]^{+}$.

5.1.10. LM-Ring fragment 16a–d. To a solution of dimethyllactone 14 (330 mg, 0.66 mmol) in dry $Et₂O$ (10 mL) was added allylmagnesium bromide (1 M in THF, 1.00 mL) at -78° C under Ar atmosphere. After stirring for 3 h, the mixture was added saturated $NH₄Cl$ solution, then warmed to room temperature, extracted with Et₂O $(X3)$. The combined organic layer was washed with brine, dried over Na₂SO₄, filtered and evaporated under reduced pressure to give a crude oil. The residue was purified by silica gel chromatography ($Et₂O$ –hexane=1:10) to give hemiacetal 15 as a colorless oil mixture (270 mg, 87% based on recovering starting material) and recover starting material (42 mg).

To a mixture of AD-mix- α (280 mg) in 50% aqueous t-BuOH (5.00 mL) was added hemiacetal 15 (90 mg) at 0° C, then the mixture was stirred for 2 days at this temperature. After starting material disappeared, the reaction was quenched by Na_2SO_3 (100 mg), then the mixture was stirred for 30 min, extracted with AcOEt $(X3)$. The combined organic layer was washed by brine, dried over $Na₂SO₄$, filtered and evaporated under reduced pressure to provide a crude oil. The residue was purified by preparative TLC $(Et₂O-hexane=1:1)$ to give LM-ring fragment 16a $(R_f=0.38, 28 \text{ mg}), \textbf{16b} (R_f=0.0.38, 27 \text{ mg}), \textbf{16c}$ $(R_f=0.0.66, 7 \text{ mg})$ and **16d** $(R_f=0.0.66, 8 \text{ mg})$ in 75% combined yield. Compound $16a$: ¹H NMR (CDCl₃, 500 MHz) δ 0.96 (3H, d, J=6.0 Hz, CH₃-59), 0.97 (3H, d, $J=6.5$ Hz, CH₃-60), 1.05 (9H, s, $-SiC(CH_3)$ ₃), 1.49 (1H, dq, $J=11.0, 6.5$ Hz, H-51), 1.54 (1H, br s, OH-54), 1.72 (1H, m, H-50), 1.94 (1H, ddd, J=14.0, 2.5, 1.0 Hz, H-53a), 2.21 $(1H, dd, J=14.0, 7.0 Hz, H=53b), 3.22 (1H, dd, J=11.0,$ 6.0 Hz, H-49), 3.70 (1H, d, $J=10.0$ Hz, H-55a), 3.88 (1H, dd, $J=11.0$, 4.0 Hz, H-47a), 4.10 (1H, d, $J=11.5$ Hz, $-CH_2Ph$, 4.11 (1H, ddd, $J=8.0, 6.0, 4.0$ Hz, H-48), 4.25 $(1H, dd, J=10.0, 4.0 Hz, H-55b), 4.26 (1H, d, J=11.5 Hz,$ $-CH_2Ph$, 4.31 (1H, dd, J=11.0, 8.5 Hz, H-47b), 7.05–7.70 (15H, m aromatic). ¹³C NMR (CDCl₃, 125 MHz) δ 13.4, 15.5, 19.2, 26.9, 34.4, 42.5, 47.0, 63.2, 71.4, 71.9, 75.2, 79.7, 109.0, 127.5, 127.59, 127.61, 127.9, 128.2, 129.5, 129.6, 133.9, 134.1, 135.7, 135.7. FAB-MS 583 [M+Na]⁺, 561 $[M+H]^+$. Compound **16b**. ¹H NMR (CDCl₃, 500 MHz) δ 1.05 (9H, s, –SiC(CH₃)₃), 1.18 (3H, d, J=7.5 Hz, CH₃-59), 1.21 (3H, d, J=7.5 Hz, CH₃-60), 1.52 (1H, br s, OH-54), 1.73 (1H, m, H-51), 1.73 (1H, dd, $J=13.5$, 4.0 Hz, H-53), 2.09 (1H, qt, $J=7.5$, 2.0 Hz, H-50), 2.39 (1H, dd,

 $J=13.5, 7.5$ Hz, H-53), 3.21 (1H, t, $J=1.5$ Hz, H-49), 3.73 $(H, dd, J=10.0, 2.0 Hz, H=55), 3.76 (1H, dd, J=10.0,$ 6.5 Hz, H-47), 3.84 (1H, dd, $J=10.0$, 6.5 Hz, H-47), 4.00 (1H, td, J=6.5, 2.0 Hz, H-48), 4.37 (1H, d, J=11.5 Hz, $-CH_2Ph$, 4.53 (1H, m, H-2), 4.65 (1H, d, J=11.5 Hz, $-CH_2P$ h), 7.23–7.69 (15H, m, aromatic). ¹³C NMR (CDCl₃, 125 MHz) ^d 18.9, 19.2, 19.8, 26.9, 35.6, 39.2, 46.9, 63.8, 69.0, 70.1, 71.0, 74.1, 110.3, 127.3, 127.6, 127.6, 127.6, 127.9, 128.2, 129.6, 133.7, 135.5, 135.6, 138.8. FAB-MS 583 [M+Na]⁺, 561 [M+H]⁺. Compound **16c**. ¹H NMR $(CDCl_3, 500 MHz)$ δ 1.02 (9H, s, $-SiC(CH_3)_3$), 1.12 (3H, d, $J=7.5$ Hz, CH₃-59), 1.20 (3H, d, $J=7.5$ Hz, CH₃-60), 1.53 (br s, 1H, OH-55), 1.59 (1H, qd, $J=7.5$, 1.0 Hz, H-51), 1.78 (1H, dd, J=13.5, 6.0 Hz, H-53a), 2.05 (1H, qt, J=7.5, 2.0 Hz, H-50), 2.17 (1H, d, $J=13.5$, 2.0 Hz, H-48), 3.05 $(1H, dd, J=2.0, 1.5 Hz, H=49), 3.62 (1H, dd, J=10.5,$ 4.5 Hz, H-47), 3.87 (1H, dd, $J=10.5$, 8.0 Hz, H-47), 3.89 $(1H, dd, J=10.0, 1.0 Hz, H=55a), 4.12 (1H, dd, J=8.0, 4.5,$ 2.0 Hz, H-48), 4.16 (1H, dd, $J=10.0$, 5.5 Hz, H-55b), 4.24 $(1H, m, H-54), 4.26$ $(1H, d, J=12.0$ Hz, $-CH₂Ph), 4.58$ $(1H,$ d, $J=12.0$ Hz, $-CH_2Ph$, 7.14–7.64 (15H, m, aromatic). ¹³C NMR (CDCl₃, 125 MHz) δ 19.0, 19.1, 19.8, 26.8, 35.1, 38.9, 45.0, 64.3, 69.2, 70.9, 71.0, 76.7, 77.5, 110.4, 127.4, 127.7, 127.7, 127.9, 128.2, 129.6, 129.7, 133.4, 133.5, 135.6, 135.6, 135.8, 138.4. FAB-MS 583 (M^{+} +Na, 5), 561 $(M^+ + H, 14)$, 543 (M⁺-OH, 6). Compound 16d. ¹H NMR $(CDCl_3, 500 MHz)$ δ 0.80 (3H, d, J=6.5 Hz, CH₃-60), 0.88 (3H, d, J=6.5 Hz, CH₃-59), 1.01 (9H, s, –SiC(CH₃)₃), 1.45 (1H, br s, OH-54), 1.52 (1H, dq, $J=10.0$, 6.5 Hz, H-51), 1.68 (1H, m, H-50), 1.86 (1H, d, $J=13.5$ Hz, H-53a), 2.02 $(1H, dd, J=13.5, 5.5 Hz, H=53b), 3.14 (1H, dd, J=10.5,$ 6.0 Hz, H-49), 3.71 (1H, dd, $J=11.5$, 4.0 Hz, H-47), 3.94 $(1H, d, J=11.5 Hz, -CH₂Ph), 3.98 (1H, dd, J=10.0, 5.0 Hz,$ H-55a), 4.03 (1H, d, $J=11.5$ Hz, $-CH_2Ph$), 4.10 (1H, ddd, $J=10.0$, 6.0, 4.0 Hz, H-48), 4.22 (1H, d, $J=10.0$ Hz, H-55b), 4.26 (1H, m, H-54), 4.27 (1H, dd, $J=11.5$, 10.0 Hz, H-47), 6.70–7.66 (15H, m, aromatic) 13C NMR $(CDCl_3, 125 MHz)$ δ 13.3, 15.4, 19.1, 26.9, 34.2, 41.8, 43.6, 62.9, 71.4, 72.2, 74.9, 78.2, 79.4, 109.7, 127.6, 127.7, 127.9, 128.2, 129.6, 129.7, 133.6, 133.9, 135.7, 135.8, 137.8. FAB-MS 583 $[M+Na]^+$, 561 $[M+H]^+$.

5.1.11. Inversion of stereogenic center at C54 of 16c and **d.** To a mixture of spiroketal $16c$ and **d** (67.4 mg) , 0.120 mmol), p-nitrobenzoic acid (90.0 mg, 0.602 mmol) and PPh₃ (158 mg, 0.602 mmol) in dry toluene (5.00 mL) was added DEAD (0.100 mL, 0.602 mmol) at room temperature under Ar atmosphere, then the mixture was stirred for 1 h. After the reaction was completed, the mixture was evaporated under reduced pressure and purified by preparative TLC (Et₂O–hexane=1:1) to give p-nitrobenzoate 16a^{\prime} (R_f =0.0.68, 26.0 mg) and 16b^{\prime} (R_f =0.0.74, 35 mg) in 95% combined yield. Compound $16a'$ (p-nitrobenzoate of **16a**) ¹H NMR (CDCl₃, 500 MHz) δ 0.97 (3H, d, J=7.5 Hz, CH_3), 1.07 (3H, d, J=6.5 Hz, CH₃), 1.07 (9H, s, $-SiC(CH_3)$ ₃), 1.55 (1H, m, H-51), 1.75 (1H, dq, J=11.0, 6.5 Hz, H-51), 2.18 (1H, d, $J=15.0$ Hz, H-53a), 2.45 (1H, dd, $J=15.0$, 7.5 Hz, H-53b), 3.24 (1H, dd, $J=11.0$, 6.0 Hz, H-49), 3.88 (1H, dd, $J=11.0$, 4.0 Hz, H-47), 4.0 (1H, d, $J=11.0$ Hz, H-55a), 4.09 (1H, d, $J=11.5$ Hz, $\text{-}CH_2\text{Ph}$), 4.14 $(1H, m, H-48), 4.23$ $(1H, d, J=11.5 Hz, -CH₂Ph), 4.48$ $(1H,$ dd, $J=11.0$, 4.5 Hz, H-55b), 5.59 (1H, br s, H-54), 7.04– 8.28 (19H, m, aromatic). ¹³C NMR (CDCl₃, 125 MHz)

^d 13.5, 15.4, 19.3, 27.0, 34.5, 42.6, 44.2, 63.2, 71.5, 72.1, 75.3, 76.6, 79.6, 108.9, 123.6, 127.6, 127.7, 127.7, 128.2, 129.6, 129.7, 130.6, 133.8, 134.1, 135.5, 135.7, 135.7, 137.9, 150.7, 164.4. Compound 16b' (p-nitrobenzoate of **16b**): ¹H NMR (CDCl₃, 500 MHz) δ 1.04 (9H, s, $-SiC(CH_3)_3$, 1.16 (3H, d, J=7.5 Hz, CH₃), 1.21 (3H, d, $J=7.5$ Hz, CH₃), 1.74 (1H, qd, $J=7.5$, 1.5 Hz, H-51), 1.95 (1H, dd, $J=14.0$, 4.5 Hz, H-53a), 2.09 (1H, qt $J=7.5$, 2.0 Hz, H-50), 2.57 (1H, dd, $J=14.0$, 7.5 Hz, H-53b), 3.21 $(1H, br s., H-49), 3.78 (1H, dd, J=10.5, 6.5 Hz, H-47), 3.84$ $(1H, dd, J=10.5, 6.0 Hz, H-47), 3.96-4.0 (2H, m, H-48,$ 55a), 4.09 (1H, dd, $J=10.5$, 5.5 Hz, H-55b), 4.36 (1H, d, $J=12.0$ Hz, $-CH_2Ph$, 4.63 (1H, d, $J=12.0$ Hz, $-CH_2Ph$), 5.50 (1H, m, H-54), 7.21–8.28 (19H, m, aromatic). 13C NMR (CDCl₃, 125 MHz) δ 18.8, 19.2, 19.7, 26.9, 35.6, 38.9, 43.6, 63.8, 69.3, 71.1, 71.2, 75.7, 76.6, 110.1, 123.5, 127.4, 127.6, 128.2, 129.6, 129.7, 130.6, 130.7, 133.7, 133.8, 135.5, 135.7, 138.7, 150.6, 164.6.

To a solution of p-nitrobenzoate $16a'$ and $16b'$ (1.00 g, 1.41 mmol) in MeOH (20 mL) was added K_2CO_3 (195 mg, 1.41 mmol) at room temperature under Ar atmosphere. After stirring for 2 h, the mixture was evaporated under reduced pressure. The residue was dissolved in distilled water and extracted with AcOEt $(X3)$. The combined organic layer was dried over $Na₂SO₄$ and filtered, evaporated under reduced pressure. The residue was purified by silica gel chromatography ($Et₂O$ –hexane=1:5) to give the mixture of 16a and b (785 mg, 99%).

5.1.12. Alcohol 25. To a solution of spiroketals 16a and b (292 mg, 0.521 mmol) in DMF (3.00 mL) was added 47.1 mg of 60% NaH (47.1 mg) at $0\degree$ C under Ar atmosphere. After stirring for 30 min, the mixture was added BnBr (269 mg, 1.56 mmol), then stirred for 5 h. The mixture was added ice-water (0.50 mL), subsequently added saturated NaHCO₃ solution, and extracted with AcOEt $(X3)$. The combined organic layer was washed with $NaHCO₃$ solution and brine, dried over $Na₂SO₄$, filtered and evaporated to give a crude oil. The residue was purified by silica gel column ($Et₂O$ –hexane=1:20) to give a mixture of benzyl ether (227 mg, 88% based on recovered starting material) and recovered starting material (70 mg). To a solution of the mixture of benzyl ether (200 mg, 0.308 mmol) in THF (5.00 mL) was added TBAF $(1.20 \text{ mL}, 1 \text{ M} \text{ in } THF, 1.20 \text{ mmol})$ at room temperature under Ar atmosphere, then the mixture was stirred for 6 h. The mixture was evaporated under reduced pressure to give a crude oil. The residue was purified by silica gel chromatography ($Et₂O$ –hexane=1:2) to give alcohol 25 $(117 \text{ mg}, 92\%)$ as colorless oil. Compound 25. ¹H NMR $(CDCl_3, 500 MHz)$ δ 0.92 (3H, d, J=6.5 Hz, CH₃), 0.96 $(3H, d, J=6.5 Hz, CH₃), 1.43$ (1H, m, H-51), 1.92 (1H, m, H-50), 2.06 (2H, ddd, $J=14.0$, 7.0, 4.0 Hz, H-53a), 3.76 (1H, m, H-49), 3.93 (4H, m, H-47, 48, 55a, 55b), 4.21 (1H, m, H-54), 4.38 (2H, d, $J=12.0$ Hz, $-CH_2Ph$), 4.47 (1H, d, $J=12.0$ Hz, $-CH_2Ph$, 4.56 (1H, d, $J=12.0$ Hz, $-CH_2Ph$), 7.24–7.34 (10H, m, aromatic). ¹³C NMR (CDCl₃, 125 MHz) ^d 13.4, 15.5, 26.5, 34.6, 42.8, 43.7, 63.1, 71.1, 72.4, 73.2, 78.2, 80.8, 109.2, 127.5, 127.7, 127.7, 128.0, 128.1, 128.4, 128.5, 134.8.

5.1.13. Chloride 26. A solution of alcohol 25 (44.2 mg,

0.107 mmol) and PPh₃ (113 mg, 0.429 mmol) in CCl₄ (10 mL) was refluxed overnight under Ar atmosphere. After the reaction was completed, the mixture was evaporated under reduced pressure and the residue was purified by preparative TLC ($Et₂O$ –hexane=1:3) to give chloride 26 $(36.6 \text{ mg}, 79\%)$. *Compound* **26**. ¹H NMR (CDCl₃, 125 MHz) δ 1.17 (3H, d, J=7.5 Hz, CH₃), 1.20 (3H, d, $J=7.0$ Hz, CH₃), 1.72 (1H, dq, $J=7.5$, 2.0 Hz, H-51), 1.85 (1H, dd, $J=13.0$, 5.0 Hz, H-53a), 2.13 (1H, tq, $J=7.5$, 2.0 Hz, H-50), 2.38 (1H, dd, J=13.0, 7.5 Hz, H-53b), 3.20 $(1H, br s, H-49), 3.58 (2H, ddd, J=11.0, 7.5, 6.0 Hz, H-47a,$ 47b), 3.92 (1H, dd, $J=10.0$, 2.5 Hz, H-55a), 4.01 (1H, dd, $J=10.0$, 6.0 Hz, H-55b), 4.04 (1H, ddd, $J=7.5$, 6.0, 2.0 Hz, H-48), 4.33, (1H, m, H-53), 4.36 (1H, d, $J=11.5$ Hz, $-CH_2Ph$), 4.44 (1H, d, J=11.5 Hz, $-CH_2Ph$), 4.67 (2H, d, J=11.5 Hz, $-CH_2Ph$), 7.24–7.34 (10H, m, aromatic). ¹³C NMR (CDCl₃, 125 MHz) δ 18.8, 19.7, 35.4, 38.8, 43.9, 44.3, 69.1, 71.1, 71.5, 71.6, 77.6, 110.5, 127.6, 127.6, 127.7, 127.8, 128.3, 128.4, 138.1, 138.3. FAB-MS 431 [M+H]⁺.

5.1.14. Iodide 27. To a mixture of alcohol 25 (54 mg, 0.129 mmol), PPh_3 (67.5 mg, 0.257 mmol) and imidazole (18 mg, 0.257 mmol) in dry toluene (5.00 mL) was added iodine (65.3 mg, 0.257 mmol) at room temperature under Ar atmosphere, then the mixture was stirred for 6 h. After the reaction was quenched by adding saturated $NaHCO₃$ solution, the mixture was extracted with $Et₂O$. The combined organic layer was washed with saturated NaHCO₃ solution and brine, dried over $Na₂SO₄$, filtered and evaporated under reduced pressure to give a crude oil. The crude oil was purified by preparative TLC to give a mixture of iodide ²⁷ (55 mg, 82%). Compound ²⁷. ¹ ¹H NMR (CDCl₃, 500 MHz) δ 0.94 (3H, d, J=7.0 Hz, CH₃), 0.97 (3H, d, J=7.5 Hz, CH₃), 2.02 (2H, m, H-51a, 53a), 2.16 (1H, dd, $J=13.5$, 7.5 Hz, H-53b), 2.23 (1H, m, H-50), 3.06 (1H, dd, $J=10.0$, 5.5 Hz, H-47a), 3.19 (1H, dd, $J=10.0$, 8.5 Hz, H-47b), 3.89 (2H, m, H-48, 55a), 4.00 (1H, dd, $J=9.5$, 5.0 Hz, H-55b), 4.22 (1H, m H-54), 4.35 (1H, d, $J=12.0$ Hz, $-CH_2Ph$), 4.38 (2H, dd, $J=12.0$ Hz, $-CH_2Ph$, 4.61 (1H, d, $J=12.0$ Hz, $-CH_2Ph$, 7.18–7.28 (10H, m, aromatic). ¹³C NMR (CDCl3, 125 MHz) ^d 6.3, 13.2, 13.3, 32.2, 34.6, 42.8, 68.9, 71.2, 71.3, 72.1, 78.1, 78.4, 110.5, 127.6, 127.6, 127.9, 128.1, 128.4, 128.4, 138.3. FAB-MS 523 [M+H]⁺, 395. Compound $27'$ (anomeric isomer of iodide 27): ¹H NMR (CDCl₃, 500 MHz) δ 1.18 (3H, d, J=7.5 Hz, CH₃), 1.22 (3H, d, J=7.5 Hz, CH₃), 1.71 (1H, qd, J=7.5, 1.5 Hz, H-51), 1.84 (1H, dd, $J=13.0$, 5.5 Hz, H-53a), 2.12 (1H, qt, $J=7.5$, 2.0 Hz, H-50), 2.40 (1H, dd, $J=13.0$, 7.5 Hz, H-53b), 3.27 (1H, t, $J=2.0$ Hz, H-49), 3.35 (1H, dd, $J=10.0$, 9.0 Hz, H-47a), 3.94 (1H, dd, $J=9.5$, 2.5 Hz, H-55a), 4.06 (1H, ddd, $J=9.0$, 5.0, 2.0 Hz, H-48), 4.17 $(1H, dd, J=9.5, 6.0 Hz, H=55b), 4.34 (1H, d, J=11.5 Hz,$ $-CH_2Ph$, 4.41 (1H, m, H-55b), 4.47 (1H, dd, J=11.5 Hz, $-CH_2Ph$, 4.70 (1H, d, J=11.5 Hz, $-CH_2Ph$), 7.24–7.34 (10H, m, aromatic). FAB-MS 523 $[M+H]^{+}$.

5.1.15. Diol 29. To a solution of tri-O-acetyl-D-glucal (200 g, 0.735 mol) and 2-propanol (112 mL, 1.47 mol, 2 equiv.) in dry CH_2Cl_2 (2.00 L) was added dropwise BF_3 ·OEt₂ (46.4 mL, 0.367 mol, 0.5 equiv.) at 0°C. After stirring for 50 min at room temperature, the reaction mixture was poured into cold saturated $NAHCO₃$ solution

and extracted with $CH₂Cl₂$. The extracts were washed with brine, dried over $Na₂SO₂$, and concentrated under reduced pressure to give acetal 28 as a crude oil, which was used in the next step without further purification.

To a solution of the crude oil of acetal 28 in MeOH (1.50 L) were added H_2O (300 mL) and Et_3N (300 mL) at room temperature. After stirring for 5.5 h at room temperature, the reaction mixture was concentrated under reduced pressure, and recrystallized (hexane/ether) to give diol 29 (117 g, 84% in 2 steps). Compound 29. Mp 98-100°C. IR (KBr) ν_{max} 3392, 2970, 2937, 2900, 1439, 1386, 1326, 1096, 1029, 947, 819, 794 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ 1.18 (3H, d, J=6.0 Hz, –CH(CH₃)₂), 1.24 (3H, d, J=6.0 Hz, $-CH(CH_3)_{2}$, 2.23 (1H, br s, $-OH$), 2.40 (1H, br s, $-OH$), 3.75 (1H, dt, $J=9.0$, 4.5 Hz, $H-48'$), 3.86 (2H, m, $H-47'a$, 47'b), 3.98 (1H, sep, J=6.0 Hz, $-CH(CH_3)_2$), 4.21 (1H, t, $J=4.5$ Hz, H-49), 5.09 (1H, br s, H-52), 5.73 (1H, dt, $J=10$, 2.5 Hz, H-50), 5.96 (1H, br-d, $J=10$ Hz, H-51). ¹³C NMR (CDCl3, 75 MHz) ^d 21.8, 23.6, 62.8, 64.3, 70.4, 71.2, 92.5, 127.0, 133.2. Anal. calcd for C9H16O4: C, 57.43; H, 8.57. Found: C, 57.26; H, 8.48.

5.1.16. Alcohol 30. To a solution of diol 29 (20.4 g, 0.108 mol) and imidazole (22.1 g, 0.324 mol, 3 equiv.) in DMF (1.00 L) was added TBSCl (19.6 g, 0.13 mol) at 0° C. After stirring for 30 min at 0° C, the reaction mixture was poured into cold saturated $NAHCO₃$ solution and extracted with $Et₂O (X3)$. The extracts were washed with brine, dried over Na₂SO₄, and concentrated under reduced pressure. The residue was purified by silica gel column chromatography $(ACOEt–hexane=1:9)$ to give alcohol 30 (32.8 g, 100%). Compound 30. IR (KBr) ν_{max} 3448, 2959, 2931, 2887, 2859, $1473, 1385, 1256, 1131, 1087, 1031, 837, 778$ cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ 0.10 (6H, s, $-Si(CH_3)_{2}$), 0.90 (9H, s, $-SiC(CH_3)$), 1.16 (3H, d, J=6.0 Hz, –CH(CH₃)₂), 1.23 (3H, d, J=6.0 Hz, –CH(CH_3)₂), 2.80 (1H, d, J=4.0 Hz, $-A$ OH), 3.77 (2H, m, H-47'a, 47'b), 3.88 (1H, m, H-48'), 3.95 $(1H, sep, J=6.0 Hz, -CH(CH₃)₂), 4.16 (1H, m, H-49), 5.04$ $(1H, m, H-52), 5.73$ $(1H, dt, J=10.5, 2.5 Hz, H-50), 5.96$ $(1H, dt, J=10.5, 1.0 Hz, H-51)$. ¹³C NMR (CDCl₃, 75 MHz) δ -5.7, -5.6, 18.2, 21.9, 23.7, 25.8, 65.5, 67.2, 69.9, 70.1, 92.4, 126.5, 132.6. Anal. calcd for $C_{15}H_{30}O_4Si$: C, 59.56; H, 10.00. Found: C, 59.57; H, 9.98.

5.1.17. Enone 31. The alcohol 30 (32.8 g, 0.108 mol) was dissolved in DMSO (600 mL) and acetic anhydride (400 mL). After stirring for 12 h at room temperature, the reaction mixture was poured into cold H_2O and extracted with $Et₂O (x3)$. The extracts were washed with brine, dried over Na₂SO₄, and concentrated under reduced pressure. The residue was purified by silica gel column chromatography $(ACOEt–hexane=1:19)$ to give enone 31 (31.6 g, 97%). Compound 31. IR (KBr) ν_{max} 2958, 2931, 2885, 2859, 1698, 1473, 1384, 1319, 1255, 1135, 1091, 1065, 1035, 917, 837, 778 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz) δ 0.05 (3H, s, $-Si(CH_3)_{2}$, 0.07 (3H, s, $-Si(CH_3)_{2}$), 0.87 (9H, s, $-SiC(CH_3)_{3}$, 1.21 (3H, d, J=6.0 Hz, $-CH(CH_3)_{2}$), 1.26 (3H, d, J=6.0 Hz, –CH(CH₃)₂), 3.98 (1H, dd, J=11.5, 5.5 Hz, H-47'a), 4.07 (1H, sep, $J=6.0$ Hz, $-CH(CH_3)_2$), 4.08 (1H, dd, J=11.5, 2.5 Hz, H-47'b), 4.49 (1H, dd, J=5.5, 2.5 Hz, H-48'), 5.40 (1H, d, J=3.5 Hz, H-36), 6.08 (1H, d, $J=10.5$ Hz, H-50), 6.84 (1H, dt, $J=10.5$, 3.5 Hz, H-51). ¹³C

NMR (CDCl₃, 75 MHz) δ -5.5, -5.5, 18.2, 21.8, 23.2, 25.7, 62.5, 71.0, 76.0, 91.4, 127.9, 144.7, 195.2. $[\alpha]_D^{29}$ = -10.32° (c 0.990, CHCl₃). Anal. calcd for $C_{15}H_{28}O_4Si$: C, 59.96; H, 9.39. Found: C, 59.96; H, 9.53.

5.1.18. Dimethylketol 33. To a stirred solution containing CuI (6.82 g, 35.8 mmol) in dry $Et₂O$ (236 mL) was added methyl lithium $(1.14 \text{ M}$ solution in Et₂O, 62.8 mL, 71.6 mmol) at 0°C. After stirring for 15 min, the reaction mixture was mixed with enone 31 (10.7 g, 35.4 mmol) with dry $Et₂O$ (118 mL). After stirring for further 30 min, to the reaction mixture were slowly added iodomethane (11.1 mL, 177 mmol) and N,N-dimethylacetamide (138 mL). The resulting mixture was gradually warmed to room temperature. After stirring 2.5 h the reaction mixture was poured into a cold 1.2N HCl solution. The resulting mixture was filtered through a Hyflo-Super-Cel[®] and extracted with Et₂O (\times 3). The extracts were dried over Na₂SO₄, and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt– hexane=1:19) to give dimethylketone 32 (11.5 g, 98%).

To a solution of dimethylketone 32 (10.8 g, 32.8 mmol) in THF (164 mL) was added TBAF (1.0 M solution in THF, 164 mL, 164 mmol) at 0° C. After stirring for 40 min at room temperature, the reaction mixture was pouring into cold saturated NH₄Cl solution, and extracted with $Et₂O (×3)$, the organic layer was dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt– hexane=1:4) to provide dimethylketol 33 (6.31 g, 89%). Compound 33. IR (KBr) ν_{max} 3436, 2974, 2367, 1729, 1637, 1457, 1381, 1071, 1011 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz), δ 1.03 (3H, d, J=6.5 Hz, CH₃-59), 1.11 (3H, d, J=6.5 Hz, CH₃), 1.15 (3H, d, J=6.0 Hz, –CH(CH₃)₂), 1.18 (3H, d, $J=6.0$ Hz, $-CH(CH_3)_{2}$, 1.58 (1H, dq, $J=12.5$, 6.5 Hz, H-51), 2.25 (1H, br, $-OH$), 2.32 (1H, dq, $J=12.5$, 6.5 Hz, H-50), 3.82 (1H, dd, J=12.0, 4.0 Hz, H-47'a), 3.91 (1H, dd, $J=12.0, 4.0$ Hz, $H-47'$ b), 3.93 (1H, sep, $J=6.0$ Hz, $-CH(CH_3)_2$, 4.17 (1H, t, J=4.0 Hz, H-48'), 4.73 (1H, d, J=6.0 Hz, H-52). ¹³C NMR (CDCl₃, 75 MHz), δ 10.2, 16.4, 21.4, 23.4, 39.9, 44.3, 62.0, 69.3, 74.5, 102.2. HRMS (FAB) calcd for $C_{11}H_{21}O_4$ [M+H]⁺ 217.1440, found 217.1463.

5.1.19. Diol 34. To a solution of dimethylketol 33 (4.89 g, 22.6 mmol) in CH_3CN (113 mL) and AcOH (113 mL) was added NaBH(OAc)₃ (16.8 g, 67.8 mmol) at -10° C. After stirring for 30 min, the reaction mixture was pouring into cold NaOH and NaHCO₃ solution, and extracted with $Et₂O$ $(\times 3)$. The organic layer was dried over Na₂SO₄ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt– hexane=1:1) to provide diol 34 (4.75 g, 96%). Compound **34.** IR (KBr) ν_{max} 3405, 2971, 2932, 1457, 1381, 1340, 1101, 1073, 1045, 1013. ¹H NMR (CDCl₃, 300 MHz), δ 1.00 (3H, d, J=7.0 Hz, CH₃), 1.07 (3H, d, J=7.0 Hz, CH₃), 1.13 (3H, d, J=6.0 Hz, –CH(CH₃)₂), 1.20 (3H, d, J=6.0 Hz, $-CH(CH_3)_2$, 1.51–1.69 (2H, m, H-51, 50), 2.19 (1H, br, $-OH$), 2.44 (1H, br, $-OH$), 3.67–3.83 (4H, m, H-49, 48', 47^{\prime} a, 47^{\prime} b), 3.88 (1H, sep, $J=6.0$ Hz, $-CH(CH_3)_2$), 4.50 (1H, d, J=3.5 Hz, H-52). [α] $_{\text{D}}^{29}$ =+110.3° (c 1.030, CHCl₃). Anal. calcd for $C_{11}H_{22}O_4$: C, 60.52; H, 10.16. Found: C, 60.41; H, 10.35.

5.1.20. Dithiane 37. To a solution of diol 34 (9.01 g, 41.3 mmol) and 1,3-propanedithiol (8.59 mL, 82.6 mmol) in CHCl3 (30.6 mL) was slowly added 12N HCl (183 mL) at 0° C. After stirring for 10 min at 0° C, the reaction mixture was poured into a cold saturated NaOH and NaHCO₃ solution slowly and extracted with AcOEt $(\times 10)$. The extracts were dried over Na₂SO₄ and concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel short column (100% AcOEt) to provide triol 35 as a dark green crude oil (15.3 g) that was used directly without further purification.

To a solution of the crude oil (15.3 g) in CH₂Cl₂ (413 mL) were added Et₃N (17.3 mL, 124 mmol), DMAP (5.04 g, 41.3 mmol) and TBSCl (7.47 g, 49.5 mmol) at room temperature. After stirring for 4 h at room temperature, the reaction mixture was poured into a cold saturated $NH₄Cl$ solution and extracted with AcOEt $(X3)$. The extracts were dried over Na₂SO₄ and concentrated under reduced pressure to provide a crude oil containing diol 36 (21.5 g) that was used directly without further purification.

To a solution of the crude oil (21.5 g) in CH₂Cl₂ (413 mL) were added 2,2-dimethoxypropane (101 mL, 825 mmol) and p-toluenesulfonic acid monohydrate (3.53 g, 18.6 mmol). After stirring 5 days at room temperature, the reaction mixture was poured into a cold saturated $NAHCO₃$ solution and extracted with $Et₂O$ (\times 3). The extracts were dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel column $(ACOE+hexane=1:9)$ to provide dithiane 37 (15.7 g, 90% in 3 steps) as a colorless oil. Compound 37. IR (KBr) ν_{max} 2934, 2360, 1464, 1380, 1250, 1220, 1102, 1073, 838, 777, 669, 517 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz) δ 0.07 (6H, s, $-Si(CH_3)_{2}$), 0.87 (3H, d, $J=6.5$ Hz, CH₃), 0.89 (9H, s, $-SiC(CH_3)_{3}$), 1.03 (3H, d, $J=6.5$ Hz, CH₃), 1.31 (3H, s, CH₃), 1.37 (3H, s, CH₃), 1.80– 1.95 (1H, m, -SCH₂CH₂CH₂S-), 2.04-2.15 (1H, m, $-SCH_2CH_2CH_2S-$), 2.29-2.46 (2H, m, H-50, 51), 2.77-2.92 (4H, m, $-SCH_2CH_2CH_2S$ –), 3.48 (1H, dd, $J=10.5$, 4.5 Hz, H-47[']a), 3.73 (1H, dd, $J=10.5$, 7.5 Hz, H-47[']b), 3.91 $(1H, dd, J=10.5, 4.5 Hz, H-49), 3.94 (1H, d, J=9.5 Hz, H-$ 52), 4.07 (1H, dt, J=7.5, 4.5 Hz, H-48[']). ¹³C NMR (CDCl₃, 75 MHz), ^d 25.6, 25.5, 10.7, 10.9, 18.2, 25.5, 25.9, 26.1, 28.4, 30.2, 30.3, 31.9, 37.2, 52.9, 62.4, 78.2, 79.5, 107.3. $[\alpha]_D^{29} = +13.76^\circ$ (c 1.015, CHCl₃). Anal. calcd for $C_{20}H_{40}O_3S_2Si$: C, 57.09; H, 9.58. Found: C, 56.98; H, 9.82.

5.1.21. Dibenzyl ether 40. A solution of dithiane $37(1.22 \text{ g})$, 2.90 mmol) in 10% HMPA/THF (19.0 mL) was treated with t-BuLi (1.48 M in pentane, 2.35 mL, 3.48 mmol) at -78° C. Immediately thereafter a precooled $(-78^{\circ}C)$ solution of (2S)-glycidylmethoxybenzyl ether (676 mg, 3.48 mmol) in 10 % HMPA/THF (9.51 mL) was added. The reaction mixture was rapidly warmed to -45° C and then quenched with saturated aqueous NH₄Cl. At ambient temperature the mixture was partitioned between $Et₂O$ and water. The organic layer was dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel column (AcOEt– hexane=1:4) to provide alcohol 38 (1.71 g, 96%).

To a solution of the alcohol 38 (1.71 g, 2.78 mmol) in THF (23.6 mL) was added tetrabutylammonium fluoride (1.0 M solution in THF, 4.17 mL, 4.17 mmol) at 0° C. After stirring for 10 min at room temperature, the reaction mixture was poured into cold saturated NH4Cl solution and extracted with Et₂O (\times 3). The organic phase was dried over Na₂SO₄ and concentrated under reduced pressure. The remaining crude oil was used directly without further purification. A one-necked flask was charged with NaH (60% dispersed in mineral oil, 455 mg, 11.4 mmol). After most of the mineral oil has been removed by washing with hexane, DMF (9.11 mL) was added to the flask, followed by a solution of the crude oil in DMF at 0° C. After stirring 1 h at room temperature, the reaction mixture was cooled to 0° C again. Benzyl bromide (0.81 mL, 6.83 mmol) and a solution of tetrabutylammonium iodide (84.0 mg, 0.278 mmol) in DMF (4.55 mL) were added to the mixture. After the addition has been complete, the mixture was allowed to room temperature and stirred for 4 h. The reaction mixture was poured into a cold saturated NH4Cl solution and extracted with Et₂O (\times 3). The extracts were dried over Na₂SO₄ and concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel column (AcOEt– hexane=1:9) to provide dibenzyl ether 40 (1.67 g, 88% in 2 steps). Compound 40. IR (KBr) ν_{max} 2933, 2361, 1616, 1514, 1457, 1379, 1248, 1074, 909, 821, 735, 698, 576, 516 cm^{-1} . ¹H NMR (CDCl₃, 300 MHz), δ 0.98 (3H, d, $J=6.5$ Hz, $-CH_3$), 1.10 (3H, d, $J=6.5$ Hz, CH₃), 1.24 (3H, s, $-OC(CH_3)_2O$ -), 1.31 (3H, s, $-OC(CH_3)_2O$ -), 1.89 (2H, m, $-SCH_2CH_2CH_2S-$), 2.15–2.30 (2H, m, H-53a, 53b), 2.58– 2.89 (6H, m, H-50, 51, $-SCH_2CH_2CH_2S$ –), 3.47 (1H, dd, $J=9.5, 6.5$ Hz, H-47'a), 3.50 (1H, dd, $J=9.5, 5.5$ Hz, H-55a), 3.55 (1H, dd, $J=9.5$, 5.0 Hz, H-55b), 3.64 (1H, dd, $J=9.5, 5.5$ Hz, H-47^{\prime}b), 3.79 (3H, s, $-C_6H_4OCH_3$), 3.90 $(1H, dd, J=10.5, 5.0 Hz, H-49), 4.04 (1H, m, H-54), 4.24$ $(1H, ddd, J=6.5, 5.5, 5.0 Hz, H-48', 4.47 (1H, d,$ $J=11.5$ Hz, $-CH_2C_6H_4OCH_3$, 4.51 (1H, d, $J=11.5$ Hz, $-CH_2C_6H_4OCH_3$), 4.54 (1H, d, J=11.0 Hz, $-OCH_2Ph$), 4.60 (2H, d, $J=11.0$ Hz, $-OCH_2Ph$), 4.69 (1H, d, $J=11.0$ Hz, $-OCH_2Ph$, 6.83–6.89 (2H, m, aromatic), 7.21–7.40 (12H, m, aromatic). ¹³C NMR (CDCl₃, 100 MHz), ^d 8.5, 13.3, 25.2, 25.8, 25.9, 28.4, 31.8, 37.3, 39.2, 55.3, 59.2, 69.8, 72.0, 72.3, 72.9, 73.5, 76.3, 76.6, 80.0, 107.6, 113.8, 127.3, 127.5, 127.8, 127.9, 128.2, 128.3, 129.2, 130.5, 138.3, 139.0, 159.2. $[\alpha]_D^{29} = -4.00^\circ$ (c 0.580, CHCl₃). Anal. calcd for C₃₉H₅₂O₆S₂: C, 68.79; H, 7.70. Found: C, 68.67; H, 7.82.

5.1.22. Aldehyde 42. A solution of dibenzyl ether 40 (11.4 g, 16.7 mmol) in 80% acetic acid (167 mL) was stirred at 40° C for 1 day. The reaction mixture was poured into a cold NaOH and NaHCO₃ solution and extracted with AcOEt (\times 4). The resulting extract was dried over Na₂SO₄, and concentrated under reduced pressure to give a crude oil that was chromatographed on a silica gel column (AcOEt– hexane=1:2) to provide diol 41 (9.89 g, 81%).

To a solution of diol 41 (9.89 g, 15.4 mmol) in dry CH_2Cl_2 (51.4 mL) was slowly added lead(IV) acetate (8.21 g) , 18.5 mmol) with dry CH_2Cl_2 (103 mL) at 0°C. After stirring for 1 h at room temperature, the reaction mixture was poured into a cold saturated $NAHCO₃$ and $Na₂SO₃$ solution and extracted with $Et₂O (x3)$. The extracts were dried over

 $Na₂SO₄$ and concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel column (AcOEt-hexane=1:4) to provide aldehyde 42 $(7.43 \text{ g}, 99\%)$ as a colorless oil. Compound 42. IR (KBr) ν_{max} 2907, 2860, 2712, 2367, 1721, 1613, 1513, 1455, 1364, 1302, 1302, 1248, 1174, 1090, 1035, 909, 821, 738, 699, 581, 522, 460 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz), δ 1.03 $(3H, d, J=7.0 \text{ Hz}, CH_3), 1.14 (3H, d, J=7.0 \text{ Hz}, CH_3), 1.90$ (2H, br m, $-SCH_2CH_2CH_2S$ -), 2.13 (1H, dd, J=15.5, 6.0 Hz, H-53a), 2.19 (1H, dd, $J=15.5$, 3.0 Hz, H-53b), 2.71– 2.76 (4H, m, $-SCH_2CH_2CH_2S-$), 2.90 (1H, qd, 7.0, 4.0 Hz, H-51), 3.05 (1H, qdd, $J=7.0$, 4.0, 1.0 Hz, H-50), 3.51 (1H, dd, $J=10.0$, 6.0 Hz, H-55a), 3.58 (1H, dd, $J=10.0$, 4.5 Hz, H-55b), 3.80 (3H, s, $-C_6H_4OCH_3$), 4.03–4.09 (1H, m, H-54), 4.49 (1H, d, J=12.0 Hz, -CH₂C₆H₄OCH₃), 4.52 (1H, d, $J=12.0$ Hz, $-CH_2C_6H_4OCH_3$, 4.63 (1H, d, $J=11.5$ Hz, $-CCH₂Ph$, 4.66 (1H, d, J=11.5 Hz, $-CH₂Ph$), 6.85–6.90 (2H, m, aromatic), 7.23–7.38 (7H, m, aromatic), 9.56 (1H, d, $J=1.0$ Hz, H-49). ¹³C NMR (CDCl₃, 75 MHz), δ 11.2, 11.6, 24.6, 25.6, 26.0, 37.3, 38.6, 47.1, 55.2, 58.4, 71.7, 71.9, 72.8, 76.0, 113.8, 127.5, 128.1, 128.3, 129.3, 130.4, 138.7, 159.3, 203.4. $[\alpha]_D^{28} = -33.87^\circ$ (c 0.555, CHCl₃). HRMS (FAB) calcd for $C_{27}H_{37}O_4S_2$ [M+H]⁺ 489.2133, found 489.2133.

5.1.23. Acetylene 47. To a solution of (trimethylsilyl)acetylene (3.22 mL, 22.8 mmol) in dry THF (76.0 mL) was slowly added t -BuLi (1.59 M) in n -hexane, 13.4 mL, 21.3 mmol) at -78° C. After stirring for 15 min at 0°C, the reaction mixture was added the solution of aldehyde 42 (7.43 g, 15.2 mmol) in dry THF (76.0 mL). After stirring for 20 min at 0° C, the reaction mixture was added iodomethane (9.46 mL, 152 mmol). After stirring for 2.5 h at room temperature, the reaction mixture was poured into a cold saturated NH₄Cl solution and extracted with Et₂O (\times 4). The extracts were dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel column $(ACOE+hexane=1:9)$ to provide a colorless oil (8.14 g, 89%) containing the propargyl ether 43.

To a solution of the propargyl ether 43 (8.01 g, 13.3 mmol) in THF (119 mL) was added TBAF (1.0 M solution in THF, 14.7 mL, 14.7 mmol) at 0° C. After stirring for 10 min at room temperature, the reaction mixture was poured into cold saturated NH₄Cl solution and extracted with Et₂O (\times 3). The organic layer was dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt– hexane=1:4) to provide the acetylene 44 (6.99 g, 99%).

To a solution of N-chlorosuccinimide (2.70 g, 20.2 mmol), silver nitrate (3.86 g, 22.7 mmol), and 2,4,6-collidine $(8.01 \text{ mL}, 60.7 \text{ mmol})$ in CH₃CN (20.2 mL) and H₂O (10.1 mL) was added a solution of the acetylene 44 $(2.54 \text{ g}, 4.80 \text{ mmol})$, in CH₃CN (20.2 mL) at -10° C. After stirring for 5 min, the reaction mixture was treated successively at 1 min intervals with saturated aqueous $Na₂SO₃$, saturated aqueous NaHCO₃, and brine (10 mL) each). The mixture was filtered through Hyflo-Super-Cel[®]. After the filter cake was washed thoroughly with 1:1 hexane– CH_2Cl_2 , the organic layer of the filtrate was dried over Na2SO4 and concentrated under reduced pressure. The

remaining crude oil was chromatographed on a silica gel column (AcOEt–hexane=1:4) to provide the a colorless oil (4.39 g) containing ketone 45 and 2,4,6-collidine.

To a solution of ketone 45 (4.39 g) in MeOH (48.0 mL) was added NaBH₄ (728 mg, 19.2 mmol) at 0 \degree C. After stirring for 30 min at 0° C, the reaction mixture was poured into cold 1.2N HCl and extracted with AcOEt $(X5)$. The organic layer was dried over Na₂SO₄ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column $(ACOE+hexane=1:9)$ to provide alcohol 46 (1.78 g, 51% in 2 steps).

To a solution of the alcohol 46 (525 mg, 1.19 mmol) in $CH₃CN$ (11.9 mL) were added pyridine (0.96 mL, 11.9 mmol) and TBSOTf $(0.55 \text{ mL}, 2.38 \text{ mmol})$ at 0°C . After stirring for 5 h at room temperature, the reaction mixture was poured into cold saturated aqueous $NH₄Cl$ and extracted with $Et₂O$ (\times 3). The extracts were washed with brine, dried over Na₂SO₄, and concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel column (AcOEt–hexane $=1:19$) to provide acetylene 47 (648 mg, 98%). Compound 47. IR (KBr) ν_{max} 3307, 2930, 2361, 1614, 1514, 1464, 1362, 1250, 1096, 836, 774, 697, 668 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz), δ 0.00–0.08 $(6H, m, -Si(CH₃)₂), 0.86–1.03$ (15H, m, $-SiC(CH₃)₃$, CH₃-59, CH3-60), 1.62–1.99 (4H, m, H-50, 51, 53a, 53b), 2.36–2.45 $(H, m, H-47), 3.32-3.40$ (3H, m, $-OCH_3$), 3.48-4.06 (5H, m, H-49, 52, 54, 55a, 55b), 3.83 (3H, s, $-OCH_2C_6H_4OCH_3$), 4.49–4.76 (4H, m, $-OCH₂Ar$), 6.90 (2H, br-d, $J=8.0$ Hz, aromatic), $7.27-7.39$ (7H, m, aromatic). ¹³C NMR (CDCl₃, 75 MHz), δ -4.5, -4.4, -4.3, -4.0, -3.9, 10.7, 10.9, 11.1, 11.5, 12.1, 12.4, 12.7, 18.0, 25.9, 34.6, 34.7, 37.5, 37.7, 38.1, 38.5, 38.6, 38.8, 38.9, 55.3, 56.5, 56.7, 71.4, 71.6, 71.9, 72.0, 72.4, 73.0, 74.1, 74.6, 74.7, 75.6, 75.8, 76.1, 81.7, 82.2, 113.8, 127.3, 127.4, 127.5, 127.7, 128.2, 129.2, 130.5, 130.6, 139.0, 139.1, 159.2. Anal. calcd for $C_{33}H_{50}O_{5}Si$: C, 71.44; H, 9.08. Found: C, 71.30; H, 9.11.

5.1.24. Pivalate 49. To a solution of methyl- α -D-glucopyranoside $(1.19 \text{ kg}, 6.13 \text{ mol})$ in N,N-dimethylformamide (6.00 L) were added 2,2-dimethoxypropane (1.88 L, 15.3 mol) and Amberlyst $15E^{\textcircled{m}}$ (6.00 g) at room temperature. After stirring for 3 days at room temperature, the reaction mixture was filtered and concentrated under reduced pressure to give the diol as gummy paste (1.43 kg) . To the solution of the paste $(100.0 \text{ g}, \text{ ca.})$ 0.42 mol) in CH_2Cl_2 (2.00 L) and pyridine (200 mL) at 0° C under N₂ atmosphere was added pivaloyl chloride (52.6 mL, 0.427 mol) dropwise. After stirring for 2 days at room temperature, the reaction mixture was poured into a cold saturated NH₄Cl solution and extracted with Et₂O (\times 3). The resulting extract was dried over $Na₂SO₄$ and concentrated in vacuo to leave a viscous oil. The oil was dissolved in $Et₂O$ containing small amount of hexane and stand still for crystallization. The mother liquors were decanted and the crystals were collected by filtration and then dried at high vacuum. The recrystallization procedure was repeated twice. A total of 92.0 g of pivaloate 49 (68%) was obtained in three crops. Compound 49. IR (KBr) ν_{max} 3476, 2990, 2917, 2880, 2840, 2361, 1736, 1482, 1374, 1270, 1198, 1167, 1042, 990, 945, 850, 750, 666, 521 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz) δ 1.20 (9H, s, -COC(CH₃)₃), 1.43 (3H,

s, $-CCH_3$, 1.52 (3H, s, $-CCH_3$), 2.92 (1H, br-d, $-OH$), 3.43 $(3H, s, -OCH_3), 3.46-3.92$ (6H, m, H-42, 43, 44, 45, 46a, 46b), 4.77 (1H, d, J=4.0 Hz, H-41). ¹³C NMR (CDCl₃, 75 MHz) ^d 18.9, 26.8, 28.9, 38.6, 55.2, 62.1, 62.7, 68.7, 73.3, 74.1, 97.5, 99.7, 178.1. $[\alpha]_D^{25} = +124.1^{\circ}$ (c 1.00, CHCl₃). Anal. calcd for C₁₅H₂₆O₇: C, 56.59; H, 8.23. Found: C, 56.59; H, 8.46.

5.1.25. Diol 54. To a solution of imidazole (17.1 g, 0.251 mol) in $CHCl₃$ (170 mL) was gradually added thiophosgene (4.80 mL, 62.8 mmol) as a solution of toluene (60.0 mL) while cooling so that temperature may not go up too much by the exothermic reaction. After the reaction mixture was stirred for 1 h at room temperature, a solution of pivaloate 49 (10.0 g) in toluene (86 mL) and CH₃Cl (20 mL) was added, and stirring the resulting mixture was continued for further 2 days at refluxing temperature. The resulting mixture was concentrated under reduced pressure to give a crude oil. The remaining oil was chromatographed on a silica gel short column (Et₂O–hexane=3:7) to provide a yellow oil containing the thiocarbamate 50 (13.1 g). NaH₂PO₂·H₂O (18.0 g, 170 mmol) and small amount of toluene were placed in a flask. After the azeotropic operation with toluene was repeated 3 times, the flask was charged with 2-methoxyethanol. (212 mL). The flask was maintained under a dry Ar atmosphere and heated at 105° C in an oil bath. The mixture was stirred vigorously and AIBN (4.18 g, 25.0 mmol) was added as a solution of 2 methoxyethanol (63.8 mL) by portions. Immediately after adding the solution of AIBN, the thiocarbamate 50 (7.28 g, 17.0 mmol) was added as a solution of 2 methoxyethanol (63.8 mL) slowly. After stirring for 10 min, the solution was poured into iced water, and the mixture was extracted with AcOEt $(X3)$. The extracts were combined and dried over $Na₂SO₄$, and the solvent was removed by evaporation on a rotary evaporator to give a yellow crude oil. The oil was chromatographed on a silica gel column ($Et₂O$ –hexane=2:1) to provide colorless oil containing the deoxygenated compound 51 (5.32 g).

To a solution of the colorless oil of 51 (32.6 g, 108 mmol) in MeOH and THF $(720 \text{ mL}$, MeOH-THF=1:1) was added NaOMe (17.5 g, 323 mmol) at room temperature. After stirring for 45 min, the reaction mixture was poured into a cold saturated NH₄Cl solution and extracted with $Et₂O (x3)$. The extracts were dried over $Na₂SO₄$, and concentrated under reduced pressure to give a crude oil that was chromatographed on a silica gel column $(Et₂O$ hexane=1:1) to provide the alcohol $\overline{52}$ (21.8 g, 93%).

To a solution of alcohol 52 (15.1 g, 69.2 mmol) in benzyl chloride (553 mL) was added potassium hydroxide (166 g) at room temperature. After stirring for 1 h at 110° C, the reaction mixture was cooled to room temperature and poured into a cooled saturated ammonium chloride solution. The resulting mixture was extracted with $Et₂O$ (\times 3). The extracts were dried over $Na₂SO₄$, and concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel short column to remove the large quantities of non-volatile benzyl alcohol generated from benzyl chloride by the aqueous work up. A yellow oil

containing the benzyl ether 53 (22.4 g) was obtained in the end, and that was used in the next step without further purification.

To a solution of the benzyl ether 53 (37.1 g, 0.12 mol) in MeOH (1150 mL) was added Amberlyst $15E^{\textcircled{w}}$ (11.5 g) at room temperature. After stirring for 40 min at room temperature, the reaction mixture was filtered and concentrated under reduced pressure to leave a viscous oil. The remaining oil was chromatographed on a silica gel short column (Et₂O–hexane=4:1) to provide diol 54 (32.0 g, 99%) in 2 steps). Compound 54. IR (KBr) ν_{max} 3324, 2907, 2361, 1734, 1456, 1378, 1330, 1237, 1182, 1106, 1052, 909, 842, 739, 697, 600, 520 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz), δ 1.84 (1H, q, $J=11.5$ Hz, H-43a), 2.16 (1H, dt, $J=11.5$, 4.5 Hz, H-43b), 3.06 (2H, br s, $-OH$), 3.39 (3H, s, $-OCH_3$), $3.42-3.63$ (3H, m, H-42, 44, 45), 3.72 (1H, d, $J=15.0$ Hz, H-46a), 3.78 (1H, d, $J=15.0$ Hz, H-46b), 4.54 (1H, d, $J=12.0$ Hz, $-OCH_2Ph$, 4.62 (1H, d, $J=12.0$ Hz, $-OCH₂Ph$), 7.24–7.38 (5H, m, aromatic). ¹³C NMR (CDCl3, 75 MHz), ^d 32.9, 54.8, 62.2, 65.6, 70.9, 72.0, 73.6, 97.0, 127.8, 127.8, 128.4, 137.9. $[\alpha]_D^{28} = +65.7^\circ$ (c 0.99, CHCl₃). Anal. calcd for $C_{14}H_{20}O_5$: C, 62.67; H, 7.51. Found: C, 62.67; H, 7.53.

5.1.26. Lactone 58. To a solution of diol 54 (30.7 g, 0.11 mol) in CH_2Cl_2 (1150 mL) were added pivaloyl chloride $(42.3 \text{ mL}, 0.34 \text{ mol})$ and DMAP $(41.9 \text{ g},$ 0.34 mol) at 0° C. After stirring for 1 h at room temperature, the reaction mixture was poured into a cold saturated $NH₄Cl$ solution and extracted with $Et₂O$ (\times 3). The extracts were dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining oil was chromatographed on a silica gel short column ($Et₂O$ –hexane=1:4) to provide the dipivaloate 55 (48.2 g, 97% in 2 steps) that was used directly in the next step without further purification.

To a solution of dipivaloate 55 (24.5 g, 56.1 mmol) in acetic anhydride (561 mL) was slowly added conc. H_2SO_4 (1.12 mL) at 0°C. After stirring for 15 min at 0°C, the reaction mixture was poured into a cold saturated $NAHCO₃$ solution and extracted with Et₂O (\times 3). The extracts were dried over Na2SO4 and concentrated under reduced pressure. The remaining oil was chromatographed on a silica gel column $(ACOEt–hexane=1:9)$ to provide the acetate 56 (22.7 g, 87%). The acetate 56 existed as a mixture of anomers.

A mixture of the acetate 56 (21.2 g, 45.6 mmol) in 1,2dimethoxyethane (414 mL), $H₂O$ (20.7 mL) and conc. HCl (20.7 mL) was stirred vigorously and heated to 55 \degree C for 14 h. After the reaction mixture was cooled to room temperature, it was poured into a cold saturated $NaHCO₃$ solution and extracted with $Et₂O$ (\times 3). The extracts were dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining oil was chromatographed on a silica gel column (AcOEt–hexane=1:4) to give the hemiacetal 57 (12.6 g, 66%) as a mixture of anomers.

The oil of the hemiacetal 57 (8.40 g, 19.9 mmol) was dissolved in DMSO (119 mL) and acetic anhydride (79.5 mL). After magnetically stirring for 14 h at room temperature, the reaction mixture was added $Et₂O$ and washed with saturated $NAHCO₃$ solution. The organic layer

was dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining oil was chromatographed on a short silica gel column (AcOEt–hexane=1:9) to give lactone 58 (7.75 g, 93%). Compound 58. IR (KBr) ν_{max} 2976, 2875, 2361, 1735, 1457, 1364, 1282, 1150, 1040, 742, 698 cm⁻¹.
¹H NMR (CDCL, 300 MHz), 8.1.20 (18H s. -COC(CH)) ¹H NMR (CDCl₃, 300 MHz), δ 1.20 (18H, s, -COC(CH₃)₃), 2.03 (1H, ddd, $J=14.5, 7.0, 6.5$ Hz, H-43a), 2.60 (1H, ddd, $J=14.5, 7.0, 6.0$ Hz, H-43b), 4.10 (1H, dd, $J=6.0, 6.5$ Hz, H-42), 4.23 (1H, dd, J=12.5, 4.0 Hz, H-46a), 4.28 (1H, dd, $J=12.5$, 2.5 Hz, H-46b), 4.65 (1H, d, $J=12.0$ Hz, $-OCH₂Ph$, 4.83 (1H, ddd, J=9.0, 4.0, 2.5 Hz, H-45), 4.87 $(1H, d, J=12.0 Hz, -OCH₂Ph), 5.04 (1H, dt, J=9.0, 7.0 Hz,$ H-44), 7.25–7.38 (5H, m, aromatic). ¹³C NMR (CDCl₃, 75 MHz), ^d 26.8, 27.0, 32.7, 38.7, 38.8, 61.9, 64.0, 71.9, 72.6, 77.5, 128.1, 128.3, 128.6, 136.8, 168.8, 177.3, 178.0. $[\alpha]_D^{28}$ =+77.9° (c 1.005, CHCl₃). Anal. calcd for C₂₃H₃₂O₇: C, 65.70; H, 7.67. Found: C, 65.84; H, 7.80.

5.1.27. Allylhydropyrane 59. To a solution of lactone 58 (14.1 g, 0.33 mmol) in THF (334 mL) was added allylmagnesium bromide $(1.0 M$ solution in Et₂O, 38.5 mL, 0.38 mmol) at -78° C. After stirring for 30 min, the reaction mixture was quenched by addition of AcOEt and poured into a cold saturated $NH₄Cl$ solution. The resulting mixture was extracted with Et₂O (\times 3) and dried over Na₂SO₄. Removal of the volatiles in vacuo gave crude product that was chromatographed on silica gel $(ACOE+hexane=3:17)$ to give hemiacetal (15.2 g) as a mixture of anomers. It was used directly in the next step without further purification. To a solution of the crude product (15.2 g, 32.9 mmol) in $CH₃CN$ (334 mL) were added $Et₃SiH$ (16.1 mL, 98.6 mmol) and BF_3 · OEt_2 (6.37 mL, 49.3 mmol) at -10° C. After stirring for 20 min, the reaction mixture was poured into a cold saturated $NAHCO₃$ solution. The resulting mixture was extracted with $Et₂O$ (\times 3) and dried over $Na₂SO₄$. Removal of the volatiles in vacuo gave crude product that was chromatographed on silica gel (AcOEt– hexane=1:19) to give allylhydropyrane 59 (8.02 g, 54% in 2 steps). Compound 59. IR (KBr) ν_{max} 3448, 3068, 2976, 2874, 2362, 1732, 1644, 1481, 1364, 1285, 1150, 1110, $1035, 990, 914, 838, 739, 699, 596$ cm⁻¹. ¹H NMR (CDCl₃, 300 MHz), δ 1.20 (18H, s, -COC(CH₃)₃), 1.44 (1H, q, $J=11.5$ Hz, H-43a), 2.24 (1H, m, H-40a), 2.61 (1H, m, H-40b), 2.72 (1H, ddd, J=11.5, 5.0, 4.0 Hz, H-43b), 3.24–3.38 (total 2H, m, H-41, 42), 3.57 (1H, ddd, $J=10.0$, 6.0, 2.0 Hz, H-45), 4.06 (1H, dd, $J=12.0$, 6.0 Hz, H-46a), 4.21 (1H, dd, $J=12.0$, 2.0 Hz, H-46b), 4.43 (1H, d, $J=11.5$ Hz, – OCH₂Ph), 4.62 (1H, d, J=11.5 Hz, $-OCH_2Ph$), 4.66 (1H, ddd, $J=11.5$, 10.0, 5.0 Hz, H-44), 4.98–5.13 (2H, m, H-57), 5.79–5.95 (1H, m, H-39), 7.25–7.38 (5H, m, aromatic). 13C NMR (CDCl₃, 75 MHz), δ 26.9, 27.0, 34.6, 35.7, 38.6, 38.7, 62.9, 66.8, 71.0, 75.0, 77.2, 80.1, 116.8, 127.8, 127.8, 128.4, 134.6, 137.9, 177.3, 178.2. $[\alpha]_D^{30} = +7.24^\circ$ (c 0.990, CHCl₃). Anal. calcd for C26H38O6: C, 69.93; H, 8.58. Found: C, 69.98; H, 8.60.

5.1.28. Alcohol 62. To a solution of allylhydropyrane 59 (8.02 g, 18.0 mmol) in MeOH (180 mL) was added NaOMe (5.82 g, 108 mmol) at room temperature. After stirring for 4 h, the reaction mixture was poured into a cold saturated NH4Cl solution. The resulting mixture was extracted with $Et₂O (X7)$ and dried over Na₂SO₄. Removal of the volatiles in vacuo gave crude product that was chromatographed on

silica gel column (AcOEt–hexane $=2:1$) to give diol 60 (4.65 g, 93%).

To a solution of diol 60 $(6.24 \text{ g}, 22.4 \text{ mmol})$ in DMF (149 mL) were added imidazole (7.63 g, 112 mmol) and TBSCl (10.1 g, 67.3 mmol) at 0° C. After stirring for 8 h at room temperature, the reaction mixture was poured into a cold saturated NH₄Cl solution and extracted with $Et₂O (×3)$. The resulting extracts were washed with H_2O (\times 2), dried over $Na₂SO₄$ and concentrated under reduced pressure to give a crude oil that was chromatographed on a silica gel short column $(ACOEt–hexane=1:19)$ to provide colorless oil containing disilylether 61 (13.0 g).

The oil of disilylether 61 was dissolved in 257 mL of methanol and treated with CSA (1.04 g, 4.48 mmol) at -10° C for 1 h. The reaction mixture was poured into a cold saturated NaHCO₃ solution. The resulting mixture was extracted with Et_2O (\times 3) and dried over Na₂SO₄. Removal of the volatiles in vacuo gave a crude product that was chromatographed on silica gel column (AcOEt– hexane=1:9) to give alcohol 62 (7.72 g, 88% in 2 steps). Compound 62. IR (KBr) ν_{max} 3496, 3074, 3032, 2930, 2859, 1642, 1473, 1456, 1362, 1253, 1096, 1005, 912, 861, 837, 777, 737, 698, 670 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz), δ 0.06 (6H, s, $-Si(CH_3)_2$), 0.88 (9H, s, $-SiC(CH_3)_3$), 1.46 (1H, q, $J=11.5$ Hz, H-43a), 2.00 (1H, br, $-OH$), 2.22 (1H, m, H-40a), 2.39 (1H, dt, $J=11.5$, 4.5 Hz, H-43b), 2.63 (1H, m, H-40b), $3.13-3.23$ (2H, m, H-42, 45), 3.31 (1H, ddd, $J=9.0$, 7.5, 3.0 Hz, H-41), 3.58 (1H, ddd, $J=11.0$, 9.0, 4.5 Hz, H-44), 3.58 (1H, br, H-46a), 3.80 (1H, br, H-46b), 4.48 (1H, d, $J=11.5$ Hz, $-OCH_2Ph$), 4.61 (1H, d, $J=11.5$ Hz, $-OCH₂Ph$, 5.00–5.13 (2H, m, H-57), 5.79–5.93 (1H, m, H-39), 7.24–7.39 (5H, m, aromatic). ¹³C NMR (CDCl₃, 75 MHz), δ -5.1, -4.3, 17.8, 25.6, 36.0, 38.9, 62.7, 66.8, 71.1, 75.5, 79.5, 81.6, 117.0, 127.9, 127.9, 128.5, 134.8, 138.2. $[\alpha]_D^{29} = +1.59^\circ$ (c 1.015, CHCl₃). Anal. calcd for $C_{22}H_{36}O_4Si$: C, 67.30; H, 9.24. Found: C, 67.26; H, 9.35.

5.1.29. Thiophenylacetylene 65. To a magnetically stirred solution containing oxalyl chloride (3.06 mL, 35.0 mmol) in $CH₂Cl₂$ (250 mL) was added a mixture of DMSO (4.97 mL, 70.1 mmol) and CH_2Cl_2 (16.7 mL) at -78° C. After stirring for 20 min, to the reaction mixture was added alcohol 62 (6.88 g, 17.5 mmol) with CH_2Cl_2 (83.5 mL). After stirring for 1 h, to the reaction mixture was slowly added Et_3N (14.7 mL, 105 mmol), which was gradually warmed to -30° C. After stirring 1 h, the reaction mixture was poured into a cooled saturated NH4Cl solution. The resulting mixture was extracted with $Et₂O$ (\times 3). The extracts were concentrated under reduced pressure. The remaining residue was passed through a silica gel and $Na₂SO₄$ short column (100% AcOEt) to provide a colorless oil containing the aldehyde 63.

To a magnetically stirred solution containing carbon tetrabromide (23.2 g, 70.1 mmol) in CH_2Cl_2 (58.4 mL) was added a solution of Ph₃P (4.46 g, 140 mmol) in CH_2Cl_2 (58.4 mL) at 0 \degree C. After stirring for 10 min at 0 \degree C, the reaction mixture was added the oil of aldehyde 63 with CH_2Cl_2 (58.4 mL). After stirring for 20 min, the reaction mixture was poured into a cooled saturated $NAHCO₃$ solution. The resulting mixture was extracted with CH_2Cl_2

 $(\times 3)$. The extracts were dried over Na₂SO₄, and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt– hexane=1:19) to give the dibromo-olefin 64 (8.76 g, 91%).

To a stirred solution containing the dibromo-olefin 64 (8.26 g, 15.1 mmol) in THF (158 mL) was added a solution of n -BuLi (21.2 mL, 33.2 mmol, 1.57 M in n -hexane) at -78° C. The reaction mixture was gradually warmed to 0°C for 1.5 h. After that, the reaction mixture was added PhSSO₂Ph (7.92 g, 31.6 mmol) with THF (63.2 mL). After stirring for 1 h, the reaction mixture was poured into cold saturated $NH₄Cl$ solution. The resulting mixture was extracted with Et₂O $(X3)$. The extracts were dried over $Na₂SO₄$, and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt–hexane= $1:19$) to give thiophenylacetylene **65** (6.71 g, 98%). Compound **65**. IR (KBr) ν_{max} 3228, 3065, 2929, 2857, 2366, 2175, 1642, 1584, 1473, 1328, 1252, 1087, 1025, 915, 838, 778, 739, 688, 595, 538 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz), δ 0.06 (6H, s, $-Si(CH_3)_2$), 0.86 $(9H, s, -SiC(CH_3)_3)$, 1.48 (1H, dt, J=12.0, 10.5 Hz, H-43a), 2.30 (1H, br-dt, $J=14.0$, 7.0 Hz, H-40a), 2.44 (1H, dt, $J=12.0$, 4.0 Hz, H-43b), 2.63 (1H, m, H-40b), 3.21–3.35 $(2H, m, H-41, 42), 3.64$ (1H, ddd, $J=11.0, 9.0, 4.5$ Hz, H-44), 4.05 (1H, d, $J=9.0$ Hz, H-45), 4.48 (1H, d, $J=11.5$ Hz, $-OCH₂Ph$, 4.61 (1H, d, J=11.5 Hz, $-OCH₂Ph$), 5.04–5.15 (2H, m, H-57), 5.86–6.01 (1H, m, H-39), 7.18–7.44 (10H, m, aromatic). ¹³C NMR (CDCl₃, 75 MHz), δ -4.7, 4.6, 17.9, 25.6, 35.9, 39.4, 70.0, 71.1, 72.6, 74.1, 74.9, 80.2, 96.9, 117.1, 126.6, 127.9, 128.5, 129.2, 132.2, 134.7, 138.1. $[\alpha]_D^{29} = +0.63^\circ$ (c 1.050, CHCl₃). Anal. calcd for C₂₉H₃₈O₃SSi: C, 70.40; H, 7.74./ Found: C, 70.41; H, 7.47.

5.1.30. Vinyl sulfone 69. To a solution of thiophenylacetylene 65 (6.71 g, 14.8 mmol) in THF (103 mL) was added TBAF (1.0 M solution in THF, 30.9 mL, 44.4 mmol) at 0° C. After stirring for 3 h at room temperature, the reaction mixture was poured into cold saturated $NH₄Cl$ solution. The resulting mixture was extracted with $Et₂O$ (\times 3). The extracts were dried over Na₂SO₄, and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column $(ACOE⁺ - hexane⁼1:4)$ to give an alcohol (4.66 g, 83%).

To a solution of the alcohol (3.43 g, 9.01 mmol) in 90.1 mL of CH_2Cl_2 were added acetic anhydride (1.27 mL, 13.5 mmol) and DMAP (1.21 g, 9.92 mmol) at 0° C. After stirring for 10 min at room temperature, the reaction mixture was poured into a cold saturated NH4Cl solution and extracted with $Et₂O$ (\times 3). The resulting extracts were dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column ($AcOE$ –hexane=1:9) to give acetate 66 (3.66 g, 96%).

To a stirred solution containing the acetate 66 (3.66 g, 8.66 mmol) in 1,2-dichloroethane (86.6 mL) were added Et₃SiH $(13.8, 86.6 \text{ mmol})$ and biscobalthexacarbonyl-2methyl-but-3-yn-2-ol (467 mg, 1.30 mmol) at room temperature. After stirring for 1.5 h at 60° C, the reaction mixture was concentrated under reduced pressure. The

remaining crude oil was chromatographed on a silica gel column (AcOEt–hexane=1:9) to give the vinyl sulfide 67 (4.30 g, 100%).

To a solution of vinyl sulfide 67 (4.30 g, 8.66 mmol) in MeOH (86.6 mL) was added K_2CO_3 (1.20 g, 8.66 mmol) at 0° C. After stirring for 1.5 h at room temperature, the reaction mixture was poured into a cold saturated NH4Cl solution and extracted with $Et₂O$ (\times 3). The resulting extracts were dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column $(ACOE-L)$ exane=3:17) to give alcohol 68 (4.17 g, 97%).

To a solution of alcohol 68 (3.77 g, 7.59 mmol) and $Na₂HPO₄$ (7.54 g, 53.1 mmol) in $CH₂Cl₂$ (75.8 mL) was added *m*CPBA (70%, 4.12 g, 167 mmol) at 0 \degree C. After stirring for 1 h, the reaction mixture was poured into a cold NaHCO₃ and Na₂SO₃ solution. The resulting mixture was extracted with Et_2O and dried over Na_2SO_4 , and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt– hexane=1:4) to give vinyl sulfone 69 (3.68 g, 85%). Compound 69. IR (KBr) ν_{max} 3484, 3067, 2956, 2876, 1603, 1447, 1298, 1237, 1140, 1082, 1004, 912, 843, 742, 699, 575 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz), δ 0.58–0.88 (15H, m, Si CH_2CH_3)₃), 1.47 (1H, q, J=11.5 Hz, H-43a), 2.16 (1H, m, H-40a), 2.55 (1H, m, H-40b), 2.73 (1H, dt, $J=12.0$, 4.5 Hz, H-43b), 3.01 (1H, d, $J=9.0$ Hz, $-OH$), 3.11–3.27 (2H, m, H-41, 42), 3.34 (1H, m, H-44), 4.45 (1H, d, $J=11.5$ Hz, $-OCH_2Ph$), 4.63 (1H, d, $J=11.5$ Hz, $-OCH_2Ph$), 4.76 (1H, t, J=9.0 Hz, H-45), 4.97–5.05 (2H, m, H-57), $5.68-5.83$ (1H, m, H-39), 6.40 (1H, d, $J=9.0$ Hz, H-46), 7.26–7.40 (5H, m, aromatic), 7.48–7.63 (3H, m, aromatic), 7.83–7.90 (2H, m, aromatic). ¹³C NMR (CDCl₃, 75 MHz), ^d 3.1, 6.9, 35.6, 39.4, 68.8, 70.8, 75.2, 78.5, 79.8, 116.7, 126.7, 127.9, 128.0, 128.5, 129.2, 133.2, 134.7, 138.1, 145.8, 154.3. $[\alpha]_D^{29} = -116^\circ$ (c 0.355, CHCl₃). Anal. calcd for $C_{29}H_{40}O_5SSi$: C, 65.87; H, 7.62. Found: C, 65.87; H, 7.73.

5.1.31. Heteroconjugate adduct 70. To a stirred solution containing acetylene 47 (2.40 g, 4.32 mmol) in dry THF (43.2 mL) was added *n*-BuLi (1.59 M) solution in hexane, 2.72 mL, 4.32 mmol) at 0° C. After stirring for 30 min, the reaction mixture was added vinyl sulfone 69 (722 mg, 1.37 mmol) with dry THF (14.4 mL). After stirring 2.5 h the reaction mixture was poured into a cooled saturated $NH₄Cl$ solution. The resulting mixture was extracted with $Et₂O$ $(X3)$. The organic layer was washed with brine, dried over Na2SO4, and concentrated under reduced pressure to give the crude oil (3.28 g) . To a solution of the crude oil (3.28 g) in THF (14.4 mL) was added TBAF fluoride (1.0 M solution in THF, 2.73 mL, 2.73 mmol) at 0° C. After stirring for 5 min at 0° C, the reaction mixture was quenched by pouring into cold saturated NH₄Cl solution, and extracted with $Et₂O$ $(X3)$. The organic layer was washed with brine, dried over Na2SO4 and concentrated under reduced pressure. The remaining crude oil that was chromatographed on a silica gel column (AcOEt–hexane=1:3) to provide heteroconjugate adduct 70 (1.05 g, 80% in 2 steps) and excess acetylene 47 (1.56 g, 65%). Compound 70. IR (KBr) ν_{max} 3448, 2930, 1613, 1514, 1455, 1306, 1250, 1087, 836, 775, 748, 698,

527 cm⁻¹. Anal. calcd for $C_{56}H_{76}O_{10}SSi$: C, 69.39 H, 7.90. Found: C, 69.39; H, 7.97.

5.1.32. Alcohol 72. To a solution of heteroconjugate adduct **70** (169 mg, 0.174 mmol) in CH₂Cl₂ (1.43 mL) was added ethyl vinyl ether (0.27 mL) and PPTS (3.0 mg) at room temperature. After stirring for 1.5 h, the reaction mixture was poured into a cold saturated $NaHCO₃$ solution and extracted with Et₂O (\times 3). The organic layer was washed with brine, dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column $(ACOE₊ - hexane=1:4)$ to give the ethoxyethyl ether 71 (181 mg, 100%).

To the solution of the ethoxyethyl ether 71 (127 mg, 0.122 mmol) in THF (1.22 mL) was added TBAF (1.0 M solution in THF, 1.22 mL, 1.22 mmol) at room temperature. After magnetically stirring for 6 days at 50° C, the reaction mixture was pouring into cold saturated NH4Cl solution, and extracted with $Et₂O (×3)$. The organic layer was washed with brine, dried over Na₂SO₄, and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column $(ACOEt–hexane=1:3)$ to provide an alcohol (99.7 mg, 88%).

To the solution of the alcohol (246 mg, 0.265 mmol) in CH_2Cl_2 (2.65 mL) were added pyridine (107 μ L, 1.33 mmol), acetic anhydride $(75 \mu L, 0.796 \text{ mmol})$ and DMAP (32.0 mg, 0.265 mmol) at room temperature. After stirring for 1 h, the reaction mixture was poured into cold saturated NH₄Cl solution, extracted with Et₂O (\times 3) and washed with brine. The organic layer was dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column $(ACOE-L)$ -hexane=1:4) to provide the acetate $(257 \text{ mg}, 100\%)$. To the solution of the acetate $(257 \text{ mg}, 100\%)$. 0.265 mmol) in MeOH (2.65 mL) was added camphorsulfonic acid (6 mg, 0.027 mmol) at room temperature. After stirring for 5 min, the reaction mixture was pouring into cold saturated NaHCO₃ solution, extracted with Et₂O (\times 3) and washed with brine. The organic layer was dried over Na2SO4 and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt–hexane=1:3) to provide alcohol 72 (218 mg, 91%). Compound 72. IR (KBr) ν_{max} 3447, 2935, 2361, 1732, 1613, 1514, 1455, 1373, 1306, 1247, 1087, 820, 749, 699, 560 cm⁻¹. Anal. calcd for $C_{52}H_{64}O_{11}S$: C, 69.62 H, 7.19. Found: C, 69.49; H, 7.29.

5.1.33. Endocyclic olefin 75 and 76. To the solution of alcohol 72 (167 mg, 0.172 mmol) in CH_2Cl_2 (0.86 mL) was added di-cobaltoctacarbonyl (295 mg, 0.862 mmol) with CH_2Cl_2 (0.86 mL) at 0°C. After stirring for 2 h at room temperature, the reaction mixture was concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel short column $(ACOE+hexane=1:1)$ to provide a crude oil (218 mg) that was used directly in the next step without further purification. To the solution of the crude oil (218 mg,) in CH_2Cl_2 (17.2 mL) was added BF_3 OEt_2 (0.22 mL, 1.72 mmol) at 0°C. After stirring for 30 min at 0° C, the reaction mixture was poured into a cold saturated NaHCO₃ solution, extracted with Et₂O (\times 3), and washed with brine. The extracts were dried over $Na₂SO₄$

and concentrated under reduced pressure. The remaining oil was chromatographed on a silica gel column (AcOEt– hexane=1:4) to give the bicyclic product 74 (165 mg, 93%).

To the solution of 74 (165 mg, 0.160 mmol) in toluene (7.99 mL) was added bis(trimethylsilyl)acetylene (1.09 mL, 4.79 mmol) and $n-Bu_3SnH$ (0.43 mL, 1.60 mmol). After stirring 30 min at 60° C, the reaction mixture was concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel column (AcOEt– hexane=9:41) to provide the endocyclic olefin 75 (81.0 mg, 68%) and its C46-epimer 76 (22.0 mg, 19%). Compound 75. IR (KBr) ν_{max} 3447, 2929, 2876, 2343, 1729, 1455, 1373, 1306, 1245, 1147, 1086, 1028, 915, 748, 699, 607, 532 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz), δ 0.87 (3H, d, $J=7.0$ Hz, CH₃), 0.88 (3H, d, $J=7.0$ Hz, CH₃), 1.43 (1H, ddd, $J=12.0$, 11.5, 11.0 Hz, H-43a), 1.55–1.60 (1H, m, H-50), 1.79–1.93 (4H, m, –OH, H-51, 53a, 53b), 1.98 (3H, s, –OCOCH3), 2.00–2.08 (1H, m, H-40a), 2.53–2.56 (1H, m, H-40b), 2.59 (1H, dt, J=12.0, 4.0 Hz, H-43b), 2.75 (1H, dd, J=9.0, 8.5 Hz, H-45), 3.05 (1H, m, H-41), 3.08 (1H, td, $J=9.0$, 2.5 Hz, H-46), 3.10 (1H, dd, $J=13.5$, 10.5 Hz, H-58a), 3.18 (1H, ddd, $J=11.0$, 9.0, 4.5 Hz, H-42), 3.34 $(1H, ddd, J=11.5, 8.5, 4.5 Hz, H-44), 3.54 (1H, m, H-54),$ 3.59 (1H, ddd, $J=11.5$, 6.5, 4.5 Hz, H-55a), 3.79 (1H, ddd, $J=11.5, 6.5, 4.5$ Hz, H-55b), 3.84 (1H, dd, $J=13.5, 1.5$ Hz, H-58b), 3.90 (1H, br, H-49), 4.42 (1H, d, $J=11.5$ Hz, $-OCH₂Ph$), 4.54 (1H, d, J=11.5 Hz, $-OCH₂Ph$), 4.58 (1H, d, $J=11.5$ Hz, $-OCH_2Ph$), 4.62 (1H, d, $J=11.5$ Hz, $-OCH₂Ph$), 4.92–5.04 (3H, m, H-52, 57), 5.62–5.73 (1H, m, H-39), 5.83 (1H, ddd, J=11.5, 4.5, 3.5 Hz, H-48), 5.93 $(1H, ddd, J=11.5, 3.5, 2.0 Hz, H-47), 7.20–7.38 (10H, m,$ aromatic), 7.52–7.56 (2H, m, aromatic), 7.62–7.66 (1H, m, aromatic), $7.88-7.91$ (2H, m, aromatic). ¹³C NMR (CDCl₃, 100 MHz), ^d 36.7, 37.0, 37.6, 37.9, 38.4, 38.5, 39.1, 39.4, 39.6, 39.7, 57.0, 57.2, 63.3, 64.2, 70.7, 70.8, 70.9, 70.9, 71.2, 72.3, 72.9, 73.0, 75.8, 75.8, 76.8, 77.1, 77.2, 77.5, 77.8, 78.9, 79.1, 79.4, 79.5, 79.6, 80.5, 81.3, 116.8, 116.9, 127.7, 127.8, 127.9, 127.9, 128.0, 128.0, 128.1, 128.2, 128.4, 128.5, 128.6, 128.9, 129.1, 131.0, 131.9, 133.6, 135.1, 135.6, 135.8, 138.1, 138.1, 138.1, 138.2, 140.1, 170.4, 170.8. Anal. calcd for C₄₃H₅₄O₉S: C, 69.14 H, 7.29. Found: C, 69.11; H, 7.25. *Compound* **76**. ¹H NMR (CDCl₃, 400 MHz), δ 0.89 (3H, d, J=7.0 Hz, CH₃), 0.90 (3H, d, $J=6.5$ Hz, CH₃), 1.38 (1H, q, $J=11.0$ Hz, H-43a), 1.43–1.51 (1H, m, H-50), 1.77–2.00 (4H, m, –OH, H-51, 53a, 53b), 2.00 (3H, s, –OCOCH3), 2.03–2.12 (1H, m, H-40a), 2.43 $(1H, m, H-40b), 2.66$ $(1H, dt, J=11.5, 4.0 Hz, H-43b), 3.05$ $(1H, dddd, J=11.0, 8.0, 4.5, 2.0 Hz, H-46), 3.09-3.16 (2H,$ m, H-41, 42), 3.16 (1H, dd, $J=9.5$, 4.5 Hz, H-45), 3.34 (1H, ddd, $J=11.5$, 9.0, 4.5 Hz, H-44), 3.40 (1H, dd, $J=14.0$, 11.0 Hz, H-58a), 3.48 (1H, dq, $J=8.0$, 4.0 Hz, H-54), 3.60 $(H, dd, J=14.0, 2.0 Hz, H-58b), 3.62 (1H, ddd, J=12.0,$ 7.5, 4.0 Hz, H-55a), 3.81 (1H, ddd, $J=12.0$, 5.0, 4.0 Hz, H-55b), 4.04 (1H, td, $J=4.0$, 2.5 Hz, H-49), 4.42 (1H, d, $J=11.5$ Hz, $-OCH_2Ph$), 4.55 (1H, d, $J=11.5$ Hz, $-OCH₂Ph$, 4.58 (1H, d, J=11.5 Hz, $-OCH₂Ph$), 4.62 (1H, d, $J=11.5$ Hz, $-OCH_2Ph$), 4.92–4.98 (2H, m, H-57), 5.08 $(H, ddd, J=10..0, 5.5, 2.5 Hz, H=52), 5.54-5.64 (1H, m,$ H-39), 5.70 (1H, dd, $J=11.5$, 4.0 Hz, H-48), 5.91 (1H, ddd, J=11.5, 8.0, 2.0 Hz, H-47), 7.25–7.38 (10H, m, aromatic), 7.51–7.58 (2H, m, aromatic), 7.61–7.65 (1H, m, aromatic), 7.88–7.91 (2H, m, aromatic). ¹³C NMR (CDCl₃, 100 MHz),

^d 11.0, 11.6, 21.2, 30.3, 35.9, 36.1, 37.3, 38.2, 39.5, 56.4, 62.8, 70.6, 71.1, 72.5, 75.1, 75.8, 78.3, 79.5, 116.8, 127.7, 127.8, 128.0, 128.1, 128.2, 128.4, 128.5, 129.2, 130.5, 133.5, 134.5, 136.0, 138.1, 138.2, 139.4, 170.4. Anal. calcd for $C_{43}H_{54}O_9S$: C, 69.14 H, 7.29. Found: C, 69.11; H, 7.25.

5.1.34. Ketone 79. To a solution of endocyclic olefin 75 (118 mg, 0.16 mmol) in THF (1.58 mL) and MeOH (1.58 mL) was treated with K_2CO_3 (39 mg, 0.28 mmol) overnight at room temperature. The reaction mixture was poured into saturated NH₄Cl solution and extracted with AcOEt $(X3)$. The organic layer was dried over Na₂SO₄ and concentrated under reduced pressure. The remaining residue was chromatographed on a silica gel column (AcOEt– hexane=2:3) to provide diol 77 (104 mg, 93%).

To a solution of the diol 77 (104 mg, 0.15 mmol) in CH_2Cl_2 (1.48 mL) were added Et₃N (206 μ L, 1.48 mmol), DMAP (18 mg, 0.15 mmol), and TBSCl (111 mg, 0.74 mmol) at 0°C. After stirring for 3 h at room temperature, the reaction mixture was poured into cold saturated NH4Cl solution, extracted with Et₂O $(X3)$, and washed with brine. The organic layer was dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt– hexane=1:4) to provide silyl ether 78 (114 mg, 94%).

To a solution of silyl ether 78 (114 mg, 0.14 mmol) in DMSO (1.39 mL) was added IBX (78 mg, 0.28 mmol) at room temperature. After stirring for 4 h, the reaction mixture was added H_2O and filtered through Hyflo-Super-Cel[®]. After the filter cake was washed with Et₂O, the organic phase of the filtrate was dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column (AcOEt– hexane=1:4) to provide ketone 79 (111 mg, 98%). Compound 79. IR (KBr) ν_{max} 3066, 2929, 2858, 1712, 1455, 1307, 1252, 1087, 837, 748, 698, 565, 533 cm⁻¹. ¹H NMR $(CDCl_3, 400 MHz), \delta 0.05$ (3H, s, $-Si(CH_3)_2)$, 0.06 (3H, s, $-Si(CH_3)_2$), 0.86 (3H, d, J=7.0 Hz, CH₃-59), 0.89 (9H, s, $-SiC(CH_3)$ ₃), 1.02 (3H, d, J=7.0 Hz, CH₃-60), 1.38 (1H, dt, J=12.0, 11.0 Hz, H-43a), 1.92-2.05 (2H, m, H-40a, 50), 2.45 (1H, dt, $J=12.0$, 4.5 Hz, H-43b), 2.49–2.56 (1H, m, H-40b), 2.63–2.81 (5H, m, H-45, 46, 51, 53a, 53b), 2.96– 3.09 (3H, m, H-41, 42, 44), 3.10 (1H, dd, $J=14.0$, 10.0 Hz, H-58a), 3.59 (1H, dd, J=10.5, 5.5 Hz, H-55a), 3.70 (1H, dd, $J=10.5$, 5.0 Hz, H-55b), 3.76 (1H, dd, $J=14.0$, 2.0 Hz, H-58b), 3.81 (1H, ddd, J=6.0, 4.0, 2.0 Hz, H-49), 4.06 (1H, dt, $J=7.5$, 4.5 Hz, H-54), 4.34 (1H, d, $J=11.5$ Hz, $-OCH₂Ph$), 4.55 (1H, d, J=11.5 Hz, $-OCH₂Ph$), 4.56 (1H, d, $J=11.0$ Hz, $-OCH_2Ph$), 4.65 (1H, d, $J=11.0$ Hz, $-OCH₂Ph$, 4.94–5.03 (2H, m, H-57), 5.61–5.74 (2H, m, H-39, 47), 5.90 (1H, ddd, $J=11.5$, 3.0, 2.5 Hz, H-48), 7.20– 7.36 (10H, m, aromatic), 7.52–7.57 (2H, m, aromatic), 7.61–7.66 (1H, m, aromatic), 7.84–7.91 (2H, m, aromatic). ¹³C NMR (CDCl₃, 100 MHz), δ 10.9, 13.7, 18.3, 25.9, 36.1, 36.8, 39.6, 44.6, 48.5, 57.3, 64.7, 70.9, 72.9, 75.0, 76.0, 78.0, 78.7, 79.3, 79.7, 116.8, 127.6, 127.7, 128.0, 128.3, 129.1, 130.6, 133.5, 134.9, 135.1, 138.1, 138.7, 140.0. Anal. calcd for $C_{47}H_{64}O_8SSi$: C, 69.08 H, 7.89. Found: C, 69.07; H, 7.87.

5.1.35. Diketone 80. To a solution of ketone 79 (71 mg, 0.087 mmol) in DMF (1.58 mL) and H_2O (158 μ L) was added PdCl₂ (1.5 mg, 8.7 μ mol) and CuCl (4.3 mg, 0.043 mmol) at room temperature. After stirring overnight under $O₂$ atmosphere, the reaction mixture was filtered through Hyflo-Super-Cel[®]. After the filter cake was washed with $Et₂O$, the organic layer of the filtrate was washed with brine, dried over $Na₂SO₄$, and concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel column $(ACOEt–hexane=3:17)$ to provide diketone 80 (62 mg, 85%). Compound 80. IR (KBr) ν_{max} 2929, 2858, 2359, 1714, 1456, 1362, 1307, 1252, 1087, 837, 749, 699, 535, 419 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz), δ 0.05 (3H, s, $-Si(CH_3)_{2}$), 0.06 (3H, s, $-Si(CH_3)_{2}$), 0.85 (3H, d, $J=7.0$ Hz, CH₃-59), 0.89 (9H, s, $-Si(CH_3)$ ₃), 1.03 (3H, d, $J=7.0$ Hz, CH₃-60), 1.40 (1H, dt, $J=12.0$, 11.0 Hz, H-43a), 1.96 (1H, qnd, J=7.0, 4.0 Hz, H-50), 2.10 (3H, s, $-COCH_3$), 2.32 (1H, dd, J=16.0, 9.5 Hz, H-40a), 2.48 (1H, dt, J=12.0, 4.5 Hz, H-43b), 2.66 (1H, q, $J=7.0$ Hz, H-51), 2.67 (1H, dd, $J=17.0$, 8.0 Hz, H-53a), 2.68 (1H, t, $J=8.0$ Hz, H-45), 2.76 (1H, dd, $J=16.0$, 3.0 Hz, H-40b), 2.78 (1H, dd, $J=17.0$, 4.0 Hz, H-53b), 2.80–2.83 (1H, m, H-46), 2.97 (1H, ddd, $J=11.0$, 9.0, 4.5 Hz, H-42), 3.05 (1H, ddd, $J=11.0$, 8.5, 4.5 Hz, H-44), 3.09 (1H, dd, $J=14.0$, 9.5 Hz, $-CH_2SO_2Ph$), 3.51 (1H, dt, $J=9.5$, 3.0 Hz, H-41), 3.57 (1H, dd, $J=14.0$, 1.5 Hz, $-CH_2SO_2Ph$), 3.59 (1H, dd, $J=10.5$, 5.5 Hz, H-55a), 3.71 (1H, dd, $J=10.5$, 5.0 Hz, H-55b), 3.82 (1H, ddd, $J=5.5$, 3.5, 1.5 Hz, H-49), 4.07 (1H, dtd, $J=7.5$, 5.5, 4.5 Hz, H-54), 4.30 (1H, d, J=11.5 Hz, -OCH₂Ph), 4.56 (1H, d, $J=11.5$ Hz, $-OCH_2Ph$), 4.57 (1H, d, $J=11.0$ Hz, $-OCH₂Ph$), 4.66 (1H, d, J=11.0 Hz, $-OCH₂Ph$), 5.67 (1H, ddd, $J=12.0$, 3.5, 2.0 Hz, H-55a), 5.81 (1H, dt, $J=12.0$, 2.5 Hz, H-47), 7.20–7.36 (10H, m, aromatic), 7.52–7.57 (2H, m, aromatic), 7.61–7.66 (1H, m, aromatic), 7.85–7.89 (2H, m, aromatic). ¹³C NMR (CDCl₃, 100 MHz), δ -5.4, 10.9, 13.9, 18.2, 25.9, 30.8, 36.7, 39.5, 39.7, 44.7, 45.8, 48.5, 57.5, 64.7, 70.6, 72.9, 75.5, 76.0, 76.2, 78.6, 78.7, 79.0, 127.6, 127.7, 127.9, 128.0, 128.3, 128.4, 129.1, 130.2, 133.5, 134.6, 137.9, 138.7, 140.1, 206.3, 212.8. HRMS (FAB) calcd for $C_{47}H_{65}O_{9}SSi$ [M+H]⁺ 833.4119, found 833.4105.

5.1.36. JKLM-ring fragment 84. To the oil of diketone 80 $(8.9 \text{ mg}, 0.011 \text{ mmol})$ were added a mixture of AD-mix- α (299 mg) and methylsulfonamide (1.0 mg, 0.011 mmol) in 50% aqueous t-BuOH (2.14 mL) in 0° C. After stirring overnight, the reaction mixture was quenched by adding $Na₂SO₃$ (320 mg), extracted with AcOEt (\times 3) and washed with saturated $NaHCO₃$ solution. The organic layer was concentrated under reduced pressure. The remaining crude oil was chromatographed on a silica gel short column (100% AcOEt) to provide an equilibrium mixture of 81 and 82 $(7.7 \text{ mg}, 83\%).$

To a solution of the equilibrium mixture of 81 and 82 $(2.9 \text{ mg}, 3.3 \text{ \mu} \text{mol})$ in CH₃CN (0.68 mL) was added two drops of HF·pyridine at room temperature. After stirring for 30 min, the reaction mixture was quenched by adding NaHCO₃ solution, extracted with Et₂O (\times 3) and washed with brine. The organic layer was concentrated under reduced pressure. The remaining crude oil was purified by preparative TLC (AcOEt–hexane=1:1) to give tetracyclic product 83 (1.8 mg, 73%).

To a solution of 83 (3.4 mg, 4.73μ mol) in MeOH

 (2.36 mL) was added large excess amount of Na₂HPO₄ (ca. 100 mg) and Hg-Na (ca. 100 mg). After stirring for 4.5 h, the reaction mixture was filtered through Hyflo-Super-Cel[®]. After the filter cake was washed with Et₂O, the organic phase of the filtrate was dried over $Na₂SO₄$ and concentrated under reduced pressure. The remaining crude oil was purified by preparative TLC $(ACOE-L)$ -hexane=1:1) to give JKLM-ring fragment 84 (2.3 mg, 82%). Compound **84**. IR (KBr) ν_{max} 3447, 2926, 1717, 1636, 1456, 1355, 1077, 1025, 938, 739, 698 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz) δ 1.01 (3H, d, J=6.5 Hz, CH₃-60), 1.06 (3H, d, $J=6.5$ Hz, CH₃-59), 1.01 (3H, d, $J=7.5$ Hz, CH₃-58), 1.40 $(1H, q, J=11.5 \text{ Hz}, H=43a), 1.48 \ (1H, dq, J=11.0, 6.5 \text{ Hz},$ H-51), 1.61 (1H, ddq, $J=11.0$, 10.0, 6.5 Hz, H-50), 2.00 $(1H, qdd, J=7.5, 5.0, 3.5 Hz, H=46), 2.08 (1H, dd, J=14.0,$ 4.0 Hz, H-53a), 2.13 (1H, dd, $J=14.0$, 6.5 Hz, H-53b), 2.14 $(3H, s, -COCH_3), 2.45$ (1H, dd, $J=15.5, 9.0$ Hz, H-40a), 2.54 (1H, dt, $J=12.0$, 4.5 Hz, H-43), 2.79 (1H, dd, $J=15.5$, 3.5 Hz, H-40b), 2.95 (1H, dd, $J=9.5$, 5.0 Hz, H-45), 3.15 $(1H, ddd, J=11.5, 9.0, 4.5 Hz, H-42), 3.26 (1H, t, J=9.5 Hz,$ H-49), 3.59 (1H, td, $J=9.0$, 3.5 Hz, H-41), 3.62 (1H, dd, $J=9.5, 2.0$ Hz, H-48), 3.65 (1H, dd, $J=3.5, 2.0$ Hz, H-47), 3.69 (1H, ddd, J=11.5, 9.5, 4.5 Hz, H-44), 3.85 (1H, dd, $J=9.5, 5.0$ Hz, H-55a), 3.96 (1H, dd, $J=9.5, 2.0$ Hz, H-55b), 4.26 (1H, dddd, $J=6.5, 5.0, 4.0, 2.0$ Hz, H-54), 4.39 (1H, d, $J=11.5$ Hz, $-OCH_2Ph$), 4.45 (1H, d, $J=12.0$ Hz, $-OCH₂Ph$), 4.47 (1H, d, J=12.0 Hz, $-OCH₂Ph$), 4.63 (1H, d, $J=11.5$ Hz, $-OCH_2Ph$), 7.20–7.37 (10H, m, aromatic). ¹³C NMR (CDCl₃, 100 MHz) δ 13.5, 15.8, 19.7, 29.7, 30.6, 36.8, 38.4, 42.0, 42.5, 46.5, 70.6, 71.1, 71.6, 71.9, 75.1, 75.8, 78.1, 78.4, 86.6, 109.3, 127.6, 127.7, 127.8, 127.8, 128.4, 138.0, 138.1, 207.2. HRMS (FAB) calcd for $C_{35}H_{47}O_8$ [M+H]⁺ 595.3271, found 595.3262.

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